

Properties of Matter

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graph TD; PM[Properties of Matter] --> PP[Physical Properties]; PM --> CP[Chemical Properties]; PP --> S[States]; PP --> CS[Changes of State]; PP --> C[Characteristics]; CP --> SC[Signs of Change]; CP --> CR[Chemical Reactions]; SC --> EX1[Examples]; CR --> EX2[Examples];
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States

Solid
Liquid
Gas
Plasma

Changes of State

Melting
Freezing
Boiling
Condensation
Ionization
Relaxation

Physical Properties

Describe a Substance

Characteristics

Color
Shape
Size
Texture
Mass
Volume
Density

Chemical Properties

Describe how it can form a new substance

Signs of Change

Odor
Change in Temp
Change in Color
Bubbles form
Solids form

Examples

Burning
Rusting
Rotting
Tarnishing

Chemical Reactions

Atoms in two substances combine to form new substances

Examples



Understanding Properties Of Matter

George Graybill



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Understanding the Properties of Matter Michael de Podesta, 1997-01-14 This introduction to the essential physics of phases of matter should be a useful textbook for introductory matter materials and solid state physics courses at degree level Its principal feature is its data driven tutorial style which enables students to familiarize themselves with practical observations on matter before tackling the theoretical rigours of the subject The book begins from first principles reminding students of their existing knowledge of theoretical physics in the context of understanding observing and measuring matter It then presents students with data of the properties of matter and through exercises and examples answers the questions what is interesting and what is there to explain about this Gases solids liquids and background theory then expanding through examining experimental data eg density thermal expansion etc Mathematical derivations are carried out at a pace appropriate to those with limited mathematical confidence and ability with only small steps between adjacent equations

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