

# Physical Science – 2nd Semester – Final Exam Study Guide – (2012–2013)

## 1. Physical Science Basics

- What tool(s) would you want to use to find the:
  - Mass of an object? triple beam balance Basic SI Unit? kg
  - Volume of an object? ruler or water displacement method Basic SI Unit? cm<sup>3</sup> or ml
  - Length of an object? Ruler Basic SI Unit? meter
  - Density of an object? graduated cylinder Basic SI Unit? ml
- What is the formula for calculating density? mass/volume
  - What is the density of a cube that is 25 g and has a volume of 3 cm<sup>3</sup>? 8g/cm<sup>3</sup>
  - Compare the density of solid H<sub>2</sub>O to liquid H<sub>2</sub>O, and explain what observable effect this has.  
solid H<sub>2</sub>O less dense than liquid H<sub>2</sub>O

## 2. Chemistry Basics

- Element (def.) substance made up of one kind of atom
- Atom (def.) basic building block of matter
- Compound (def.) a molecule that contains atoms or molecules of two or more different elements
- Molecule (def.) a combination of two or more of the same or different atoms that are bonded together

## 3. Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that many scientists throughout history have created models to describe them.

Why have atomic theories changed over time? Better technology came around.

| Scientist         | Year | Major Discovery/ Contribution                                                                                                                                                                                           | Name of atomic model |
|-------------------|------|-------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|----------------------|
| John Dalton       | 1808 | Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different masses.                                                                                                | Billiard Ball Model  |
| J.J. Thomson      | 1897 | Negatively charged electron                                                                                                                                                                                             | Plum Pudding Model   |
| Niels Bohr        | 1913 | Electrons move <u>fixed orbits</u> around the atom's nucleus in specific layers, or <u>electron shells</u> . Atoms absorb or give off energy when the electrons move from one shell ( <u>energy level</u> ) to another. | Bohr Model           |
| Ernest Rutherford | 1909 | 1909 /Based on idea that positive reflects positive/ atom mostly empty space; electrons orbit randomly around positively charged nucleus                                                                                | Planetary Model      |

## 4. Periodic Table

- The current periodic table is arranged by increasing atomic number.
- Reading the Periodic Table
  - An element's properties can be predicted from its location in the periodic table.
  - The atomic number represents the number of protons.
  - The atomic mass represents the number of protons and neutrons.

# Study Guide For Physical Science Final Exam

**DP Hallahan**



## Study Guide For Physical Science Final Exam:

Physics Handbook & Study Guide Grade 11-12 IEB Kevin Smith,2024-02-01 A comprehensive summary of Grade 11 12 Physics Simple logical summaries with example exam questions and work through solutions The book covers the fundamentals of Grade 11 12 Physics and complements the material in any class text X-kit Exam 2004 Physical Science ,2005 *Principles of Physical Science I DANTES/DSST Test Study Guide - PassYourClass* PassYourClass,2011-03-01 Our DANTES study guides are different The Principles of Physical Science1 DANTES DSST study guide TEACHES you everything that you need to know to pass the DSST test This study guide is more than just pages of sample test questions Our easy to understand study guide will TEACH you the information We ve condensed what you need to know into a manageable book one that will leave you completely prepared to tackle the test This study guide includes sample test questions that will test your knowledge AND teach you new material Your Principles of Physical Science 1 study guide also includes flashcards Use these to memorize key concepts and terms Anyone can take and pass a DANTES test What are you waiting for Resources in Education ,2001-10 The Chemical News and Journal of Physical Science ,1905 **Chemical News and Journal of Physical Science** ,1887 **Research in Education** ,1968 *A Summary of Research in Science Education* ,1974 *Computer Supported Collaborative Learning 2005* Timothy Koschmann,2017-10-03 The Computer Supported Collaborative Learning CSCL conference has become an internationally recognized forum for the exchange of research findings related to learning in the context of collaborative activity and the exploration of how such learning might be augmented through technology This text is the proceedings from CSCL 2005 held in Taipei Taiwan This conference marked the 10th anniversary of the first CSCL Conference held at Indiana University in 1995 Subsequent meetings have been held at the University of Toronto Stanford University University of Maastricht Netherlands University of Colorado at Boulder and the University of Bergen Norway Just as the first CSCL conference was instrumental in shaping the trajectory of the field in its first decade the conference in Taipei will play an important role in consolidating an increasingly international and interdisciplinary community and defining the direction of the field for the next 10 years This volume and the papers from which it is comprised will be an important resource for those active in this area of research and for others interested in fostering learning in settings of collaboration **The popular educator** Popular educator,1884 *Books and Pamphlets, Including Serials and Contributions to Periodicals* Library of Congress. Copyright Office,1968 **Trends in Education** ,1975 Catalog of Copyright Entries. Third Series Library of Congress. Copyright Office,1968 Includes Part 1 Number 2 Books and Pamphlets Including Serials and Contributions to Periodicals July December *Science physics papers. Being the questions for 21 years [1869-1889], (the questions from 1875 to 1896, to 1899, to 1900, to 1903, 1880 to 1905, to 1906, to 1907, 1885 to 1908, to 1909, to 1910, to 1911).* London univ, exam. papers,1889 **Team-Based Learning** Larry K. Michaelsen,Arletta Bauman Knight,L. Dee Fink,2023-07-03 This book describes team based learning TBL an unusually

powerful and versatile teaching strategy that enables teachers to take small group learning to a whole new level of effectiveness. It is the only pedagogical use of small groups that is based on a recognition of the critical difference between groups and teams and intentionally employs specific procedures to transform newly formed groups into high performance learning teams. This book is a complete guide to implementing TBL in a way that will promote the deep learning all teachers strive for. This is a teaching strategy that promotes critical thinking, collaboration, mastery of discipline knowledge, and the ability to apply it. Part I covers the basics, beginning with an analysis of the relative merits and limitations of small groups and teams. It then sets out the processes with much practical advice for transforming small groups into cohesive teams for creating effective assignments and thinking through the implications of team based learning. In Part II, teachers from disciplines as varied as accounting, biology, business, ecology, chemistry, health education, and law describe their use of team based learning. They also demonstrate how this teaching strategy can be applied equally effectively in environments such as large classes, mixed traditional and on line classes, and with highly diverse student populations. Part III offers a synopsis of the major lessons to be learned from the experiences of the teachers who have used TBL as described in Part II. For teachers contemplating the use of TBL, this section provides answers to key questions, e.g., whether to use team based learning, what it takes to make it work effectively, and what benefits one can expect from it for the teacher as well as for the learners. The appendices answer frequently asked questions, include useful forms and exercises, and offer advice on peer evaluations and grading. A related Web site that allows readers to continue the conversation, view video material, access indexed descriptions of applications in various disciplines, and post questions further enriches the book. The editors claim that team based instruction can transform the quality of student learning; this is fully supported by the empirical evidence and examples they present. An important book for all teachers in higher education. *Research in Education*, 1972. **Physical Sciences**

National Learning Corporation, 2020-03-15. **Otto E. Miller, Plaintiff-Respondent, Against Fred W. Smythe, Defendant-Appellant**, Directory of Distance Learning Opportunities Modoc Press, Inc., 2003-02-28. This book provides an overview of current K-12 courses and programs offered in the United States as correspondence study or via such electronic delivery systems as satellite cable or the Internet. The Directory includes over 6,000 courses offered by 154 institutions or distance learning consortium members. Following an introduction that describes existing practices and delivery methods, the Directory offers three indexes: Subject Index of Courses Offered by Level, Course Level Index, Geographic Index. All information was supplied by the institutions. Entries include current contact information, a description of the institution, and the courses offered, grade level, and admission information, tuition and fee information, enrollment periods, delivery information, equipment requirements, credit and grading information, library services, and accreditation. **An Introduction to Physical Science** James T. Shipman, Jerry D. Wilson, Aaron W. Todd, 2002-07-08. This text presents a survey of the physical sciences: physics, chemistry, astronomy, meteorology, and geology for non-science majors. Topics are treated

both descriptively and quantitatively providing flexibility for instructors who wish to emphasize a highly descriptive approach a highly quantitative approach or any spectrum in between

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