

Name: _____ Date: _____ Class: _____

Solubility Curve Practice Problems Worksheet 1



You'll notice that for most substances, solubility increases as temperature increases. As discussed earlier in solutions involving liquids and solids typically more solute can be dissolved at higher temperatures. Can you find any exceptions on the graph? _____

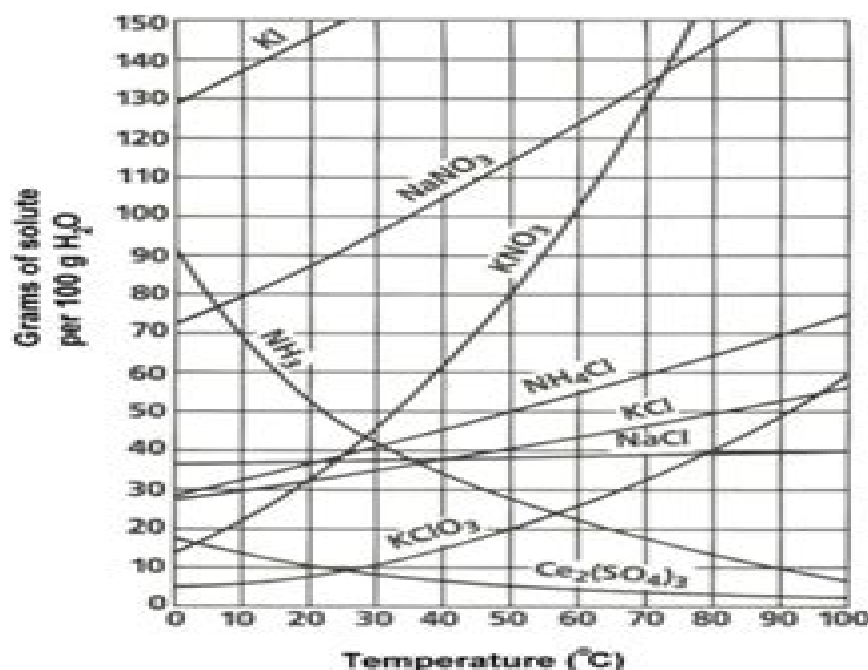
Here's an example of how to read the graph. Find the curve for KClO_3 .

At 30°C approximately 10g of KClO_3 will dissolve in 100g of water. If the temperature is increased to 80°C , approximately _____ of the substance will dissolve in 100g (or 100mL) of water.

Directions: Use the graph to answer the following questions. REMEMBER UNITS!

1) What mass of solute will dissolve in 100mL of water at the following temperatures?

- KNO_3 at 70°C = _____
- NaCl at 100°C = _____
- NH_4Cl at 90°C = _____
- Which of the above three substances is most soluble in water at 15°C . = _____



Solubility Curve Practice

Kaplan Test Prep

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the Kaplan Test Prep text.

Solubility Curve Practice:

Practical Chemical Thermodynamics for Geoscientists Bruce Fegley, Rose Osborne, 2013 Presents brief historical summaries and biographies of key thermodynamics scientists alongside the fundamentals they were responsible for

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Plant Engineering

Practice F. W. Dodge Corporation, Plant Engineering, 1958 **Cracking the GED Test with 2 Practice Tests, 2017 Edition** Princeton Review, 2016-07 Two full length practice tests included **MASS TRANSFER** N. Anantharaman, K. M. Meera Sheriffa Begum, 2011-05 Mass transfer operations are of great importance in a process industry as it has a direct impact on the cost of the final product A chemical process engineer therefore should have sound knowledge of the basics of mass transfer and its applications This book is designed to equip the reader with sufficient knowledge of mass transfer operations and face the challenges ahead The objective of this textbook is to teach a budding chemical engineer the principles involved in analyzing a process and apply the desired mass transfer operation to separate the components involved It deals with operations involving diffusion interphase mass transfer humidification drying crystallization absorption distillation extraction leaching and adsorption The principles and equipment used for different mass transfer operations have been lucidly explained Designed for a two semester course this text is primarily intended for the undergraduate students of chemical pharmaceutical petrochemical engineering as well as biotechnology and industrial biotechnology It will also be useful to plant engineers and design professionals **KEY FEATURES** 1 Explains the theoretical concepts with full derivation of equations 2 Illustrates the application of theory through worked out numerical examples 3 Provides exercise problems with answers at the end of each chapter for practice Nutrition Support Practice Manual ,2001-11 *Microemulsions Theory and Practice* Leon Prince, 2012-12-02 Microemulsions Theory and Practice covers the development of the theory and practice of microemulsion systems This book is divided into seven chapters that explore the physics and chemistry of microemulsions This book deals first with the commercial history of microemulsions from the discovery of carnauba wax emulsions to polymer emulsions This topic is followed by discussions on the theoretical aspects of microemulsion formulation techniques and the design of other products The subsequent chapter describes the microemulsion formulation with less solubilizer or emulsifier together with their optical properties A chapter examines the mixed film theory that explains the dispersions oil water interface and inferences in microemulsions Another chapter considers the role of microemulsions in micellar solutions and their relations to the concentrations of different compounds This chapter also looks into the association phenomena of three component phase equilibria diagrams and liquids crystals to microemulsions The concluding chapter discusses the role of the capillary and hydrostatic forces on the entrapment of oil in the reservoir and the necessary conditions for the displacement of entrapped oil The important properties and economic aspects of a microemulsion slug required for the tertiary oil recovery are also covered in this chapter Journal of the Chemical Society Chemical Society (Great Britain), 1914 Titles of chemical papers in British and foreign journals included in Quarterly journal v 1 12 **Journal of the Chemical Society** ,1914

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