

| <u>Ion</u>         | <u>Solubility</u> | <u>Exceptions</u>                                                                                                      |
|--------------------|-------------------|------------------------------------------------------------------------------------------------------------------------|
| $\text{NO}_3^-$    | soluble           | none                                                                                                                   |
| $\text{ClO}_4^-$   | soluble           | none                                                                                                                   |
| $\text{Cl}^-$      | soluble           | except $\text{Ag}^+$ , $\text{Hg}_2^{2+}$ , * $\text{Pb}^{2+}$                                                         |
| $\text{I}^-$       | soluble           | except $\text{Ag}^+$ , $\text{Hg}_2^{2+}$ , $\text{Pb}^{2+}$                                                           |
| $\text{SO}_4^{2-}$ | soluble           | except $\text{Ca}^{2+}$ , $\text{Ba}^{2+}$ , $\text{Sr}^{2+}$ ,<br>$\text{Hg}^{2+}$ , $\text{Pb}^{2+}$ , $\text{Ag}^+$ |
| $\text{CO}_3^{2-}$ | insoluble         | except Group IA and $\text{NH}_4^+$                                                                                    |
| $\text{PO}_4^{3-}$ | insoluble         | except Group IA and $\text{NH}_4^+$                                                                                    |
| $\text{-OH}$       | insoluble         | except Group IA, * $\text{Ca}^{2+}$ ,<br>$\text{Ba}^{2+}$ , $\text{Sr}^{2+}$                                           |
| $\text{S}^{2-}$    | insoluble         | except Group IA, IIA and<br>$\text{NH}_4^+$                                                                            |
| $\text{Na}^+$      | soluble           | none                                                                                                                   |
| $\text{NH}_4^+$    | soluble           | none                                                                                                                   |
| $\text{K}^+$       | soluble           | none                                                                                                                   |

\*slightly soluble

# Solubility Rules And Net Ionic Equations Pogil Activity Key

**Lifelique**



**Solubility Rules And Net Ionic Equations Pogil Activity Key:**

Net Ionic Equations Lifeliqe,2019 This lesson plan covers the various ways to depict chemical equations using net ionic equations

## **Solubility Rules And Net Ionic Equations Pogil Activity Key** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the ability of words has become more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such is the essence of the book **Solubility Rules And Net Ionic Equations Pogil Activity Key**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book's key themes, examine its writing style, and analyze its overall affect readers.

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