

Date _____
Course/Section Chem 111-03L
Instructor _____

Student Name Tre' Carr
Team Members _____

95⁷³

Solubility Within a Family

Pre-laboratory Assignment

1. Give the names and symbols of the alkaline earth metals that you will encounter in this experiment.

Mg - Magnesium ⁺²

Ca - Calcium ⁺²

Sr - Strontium ⁺²

Ba - Barium ⁺²

2. a. What is the general electron configuration of the alkaline earth metals?
normally ends with a s^2

- b. What is the electron configuration of lead?

$1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6s^2 6p^2$

- c. How do the formulas of the oxides formed by the alkaline earth metals compare with those of lead?

The formulas of the oxides formed by the alkaline earth metals form oxides with the same general formula, but with lead they will form nitrates, hydroxides, chlorides, bromides, iodides, sulfates, carbonates, oxalates, and chromates.



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Ian Pickup



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