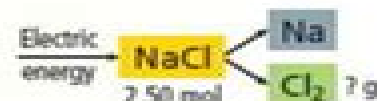


## PRACTICE Problems

Do additional problems...

Online Pro

13. Sodium chloride is decomposed into the elements sodium and chlorine by means of electrical energy. How much chlorine gas, in grams, is obtained from the process diagrammed at right?



14. **Challenge** Titanium is a transition metal used in many alloys because it is extremely strong and lightweight. Titanium tetrachloride (TiCl<sub>4</sub>) is extracted from titanium oxide (TiO<sub>2</sub>) using chlorine and coke (carbon).



- What mass of Cl<sub>2</sub> gas is needed to react with 1.25 mol of TiO<sub>2</sub>?
- What mass of C is needed to react with 1.25 mol of TiO<sub>2</sub>?
- What is the mass of all of the products formed by reaction with 1.25 mol of TiO<sub>2</sub>?

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# Study Guide For Content Mastery Section 12 2

## Stoichiometric Calculations

**Xiaolong Qi**



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