

Unit 10



UNIT 10: PERIODIC TABLE: CONCEPTS AND APPLICATIONS

Section 10.1: Introduction

In your notebook, read aloud the modern periodic table.

Copy the following table in the form on the left below from the periodic table to complete the table on the right.

1
2
3
4
5
6
7

Atomic Mass	1.1
Atomic Number	1.2
Electron Configuration	1.3
Chemical Name	1.4
Chemical Symbol	1.5

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 1. It is located in the periodic table.
- _____ 2. It is the sum of the periodic table.
- _____ 3. Group 1 elements.
- _____ 4. Elements that are strong oxidizing agents.
- _____ 5. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 1. There are two representative elements of elements.
- _____ 2. There are two groups of the elements in the periodic table.
- _____ 3. Group 1 elements are highly reactive elements in the periodic table.
- _____ 4. Group 17 elements are the most reactive elements.
- _____ 5. Group 1 elements are highly reactive elements in the periodic table.
- _____ 6. Group 17 elements are highly reactive elements in the periodic table.
- _____ 7. Group 1 elements are highly reactive elements in the periodic table.
- _____ 8. Group 17 elements are highly reactive elements in the periodic table.

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Scott C. Dulebohn



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