

Now _____ Date _____ Class _____

CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

Covalent Bonding

Section 9.1 The Covalent Bond

In your textbook, read about the nature of covalent bonds.

Use each of the terms below just once to complete the passage.

covalent bond	molecule	sigma bond	exothermic	p bond
---------------	----------	------------	------------	--------

When sharing of electron occurs, the attachment between atoms that results is called a(n) **(1) covalent bond**. When such an attachment is formed, bond dissociation energy is released, and the process is **(2) exothermic**. When two or more atoms bond by means of electron sharing, the resulting particle is called a(n) **(3) molecule**. If the electrons shared are contained between the two atoms, the attachment is called a(n) **(4) sigma bond**. If the sharing involves the overlap of parallel orbitals, the attachment is called a(n) **(5) p bond**.

In your textbook, read about single and multiple bonds and bond strengths.

Circle the letter of the choice that best completes the statement or answers the question.

- Is what form the elements such as hydrogen, nitrogen, and oxygen normally exist?
 - as single atoms
 - (b) as molecules containing two atoms**
 - as molecules containing three atoms
 - as molecules containing four atoms
- How many electrons are shared in a double covalent bond?
 - none
 - one
 - two
 - (d) four**
- Bond length is the distance between
 - two molecules of the same substance
 - (b) the nucleus of two attached atoms**
 - the electron in two attached atoms
 - the orbitals of two attached atoms
- Which of the following relationships relating to bond length is generally correct?
 - (a) the shorter the bond, the stronger the bond**
 - the shorter the bond, the weaker the bond
 - the shorter the bond, the lower the electron in it
 - the shorter the bond, the lower the bond dissociation energy

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Chemistry: Matter and Change • Chapter 9

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Now _____ Date _____ Class _____

CHAPTER 9 STUDY GUIDE FOR CONTENT MASTERY

Section 9.2 Naming Molecules

In your textbook, read about how binary compounds and acids are named from their formulas.

For each statement below, write true or false.

- Binary molecular compounds are generally composed of a metal and a nonmetal. **false**
- The second element in the formula of a binary compound is named using the suffix -ite. **false**
- The prefix zero indicates three atoms. **false**
- The prefix four indicates six atoms. **true**
- In naming the first element in a formula, the prefix mono- is not used. **true**
- For binary acids, the hydrogen part of the compound is named using the prefix hydro-. **true**
- An oxyacid contains only two elements. **false**
- If the name of the anion of an oxyacid ends in -ite, the acid name contains the suffix -ous. **false**

In your textbook, read about naming molecular compounds and oxyacids.

For each item in Column A, write the letter of the matching item in Column B.

Column A

- (c)** N_2O
- (d)** H_2CO_3
- (g)** H_2SO_4
- (h)** N_2O_5
- (f)** HNO_3
- (i)** HNO_2
- (b)** HBr
- (e)** HBrO_3

Column B

- hydrobromic acid
- nitrogen trioxide
- carbon monoxide
- nitrous acid
- nitric acid
- nitrous acid
- nitric acid
- carbon dioxide

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Daniel C. Pantaleo



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Kenneth J. Hughes, Paul B. Kelter, 2002-05 Reviews essential points from the text presents problems and hints to assist the student in solving them Solutions to selected problems in the end of chapter problem sets are included By Kenneth Hughes of the University of Wisconsin Madison Study Guide [to Accompany] General Chemistry James E. Brady, 1982

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web nov 6 2023 example 1 student sick leave for short term illness subject sick leave for student s name short term illness dear recipient s name i am writing to inform you that my child student s name is down with a high fever and has been

advised rest for number of days

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web nov 8 2023 example 1 subject sick leave application your name dear recipient s name i regret to inform you that i m not in a condition to attend class due to a sudden bout of illness i kindly request a leave of absence from start date to end date i will ensure to catch up with the missed lessons upon my return thank you for your understanding

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web mar 10 2023 sample 1 leave of absence letter for school dear mr burr i am writing to request a 3 day leave of absence for christopher from school as we have arranged to attend a wedding at the other side of the country and need considerable time to

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web sample sick leave applications for students to get leave against fever temperature accident surgery flu cough infection or any type of injury from school college or university application for fever by student in english application for leave in school by parents application to authority informing them regarding late joining due to

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web mar 2 2023 whether you are in school or college or are a working professional in case of informing the concerned authorities about your absence due to illness you are required to write an application for sick leave therefore knowing the correct format for a sick leave mail or application is very important for effective communication with the

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web sample school leave application for fever to the principal st xaviers high school mumbai india date dd mm yyyy subject application requesting leave for fever respected sir madam with due respect and humble submission i beg to state to you that i am a regular student of your esteemed school i have high fever and am unable to

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