

10.1 Periodic Trends



10.1 PERIODIC TRENDS FROM CONCEPTUAL MASTERY

Section 10.1 continued

In your notebook, read aloud the student periodicals table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

Atomic Mass	_____
Atomic Number	_____
Electron Configuration	_____
Chemical Name	_____
Chemical Symbol	_____

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 16. It increases in the periodic table.
- _____ 17. It decreases in the periodic table.
- _____ 18. Group 1 elements.
- _____ 19. Elements that are shiny and conduct electricity.
- _____ 20. Group 17 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write *yes* if the statement is true, if the statement is false, change the underlined word or phrase to make it true.

- _____ 21. There are two representative elements of elements.
- _____ 22. Most elements are located in the periodic table are metals.
- _____ 23. Group 1A elements except for hydrogen are located in the alkali metals.
- _____ 24. Group 2A elements are the alkaline earth metals.
- _____ 25. Group 1A elements are highly reactive elements known as halogens.
- _____ 26. Group 1A elements are very reactive elements known as representative elements.
- _____ 27. Alkaloids have properties of both metals and non-metals.

Study Guide For Content Mastery Stoichiometric

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