

Mixtures, Solubility, and Acid/Base Solutions

Substances and Mixtures

..... Before You Read

What do you think? Read the two statements below and decide whether you agree or disagree with them. Place an A in the Before column if you agree with the statement or a D if you disagree. After you've read this lesson, reread the statements to see if you have changed your mind.

Before	Statement	After
	1. You can identify a mixture by looking at it without magnification.	
	2. A solution is another name for a homogeneous mixture.	

..... Read to Learn

Matter: Substances and Mixtures

Many different types of matter exist around you. On your way to school, you might see metal, plastic, rocks, concrete, plants, fabric, water, and skin. You might notice that you can group some types of matter together into one category. For example, keys, coins, and paper clips are made of metal. Grouping matter into categories helps you understand how some things are similar to each other, but different from other things. Nearly all types of matter can be sorted into just two major categories—substances and mixtures.

What is a substance?

A **substance** is matter that is always made up of the same combination of atoms. There are two types of substances—elements and compounds. Recall that an element is matter made of only one type of atom, such as carbon (C), oxygen (O_2), and chlorine (Cl_2). A compound is matter made of atoms of two or more elements that are chemically bonded together. For example, water (H_2O) always contains two atoms of hydrogen bonded to one atom of oxygen. Table salt ($NaCl$) always contains one atom of sodium bonded to one atom of chlorine. The composition of elements and compounds doesn't change. Therefore, all elements and compounds are substances. 🔍

Key Concepts 🔑

- How do substances and mixtures differ?
- How do solutions compare and contrast with heterogeneous mixtures?
- In what three ways do compounds differ from mixtures?

Study Coach

Make a Table with two columns to contrast substances and mixtures. Label one column *Substances* and the other column *Mixtures*. Complete the table as you read this lesson.

✓ Reading Check

1. Analyze Is table salt an element or a compound? Explain.

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Y. Samiullah



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Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

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