

Study Guide

Key Concepts

- Concentrations can be measured qualitatively and quantitatively.
- Molarity is the number of moles of solute dissolved per liter of solution.

$$\text{molarity (M)} = \frac{\text{moles of solute}}{\text{liters of solution}}$$

- Molality is the ratio of the number of moles of solute dissolved in 1 kg of solvent.

$$\text{molality (m)} = \frac{\text{moles of solute}}{\text{kg of solvent}}$$



Solution Concentration Study Guide Answers

Lois A Ball



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time formula $p = \frac{w}{t}$ where p power w work done t time taken definition of work energy theorem the work done on a body by applying force is equal to the change in kinetic energy of the body this is defined as work

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