

Name: _____ Date: _____ Class: _____

Solubility Curve Practice Problems Worksheet 1



You'll notice that for most substances, solubility increases as temperature increases. As discussed earlier in solutions involving liquids and solids typically more solute can be dissolved at higher temperatures. Can you find any exceptions on the graph? _____

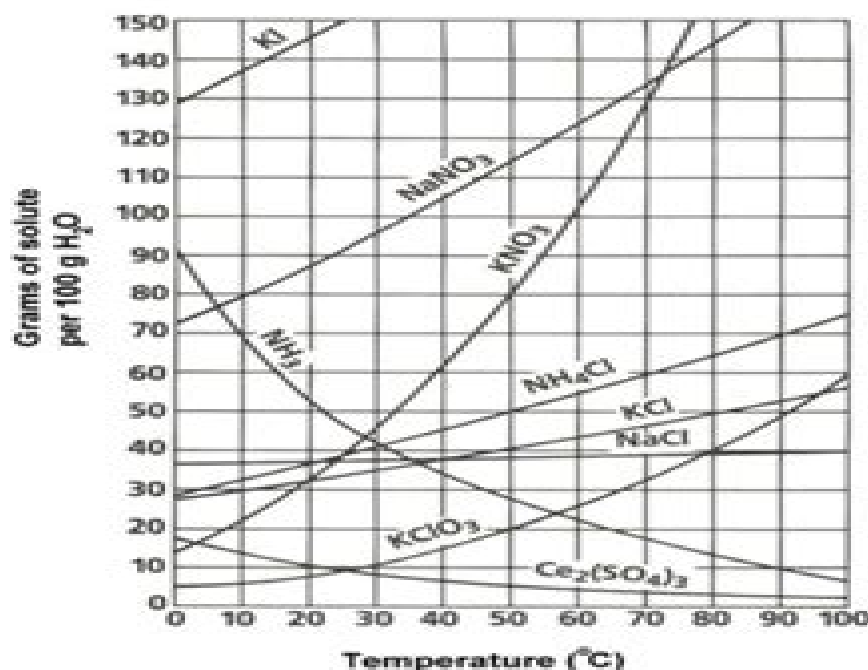
Here's an example of how to read the graph. Find the curve for KClO_3 .

At 30°C approximately 10g of KClO_3 will dissolve in 100g of water. If the temperature is increased to 80°C , approximately _____ of the substance will dissolve in 100g (or 100mL) of water.

Directions: Use the graph to answer the following questions. REMEMBER UNITS!

1) What mass of solute will dissolve in 100mL of water at the following temperatures?

- KNO_3 at 70°C = _____
- NaCl at 100°C = _____
- NH_4Cl at 90°C = _____
- Which of the above three substances is most soluble in water at 15°C . = _____



Solubility Curve Practice Problems With Answers

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 - Chapter 2: Essential Elements of Solubility Curve Practice Problems With Answers
 - Chapter 3: Solubility Curve Practice Problems With Answers in Everyday Life
 - Chapter 4: Solubility Curve Practice Problems With Answers in Specific Contexts
 - Chapter 5: Conclusion
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 3. In chapter 2, the author will delve into the foundational concepts of Solubility Curve Practice Problems With Answers. The second chapter will elucidate the essential principles that need to be understood to grasp Solubility Curve Practice Problems With Answers in its entirety.
 4. In chapter 3, the author will examine the practical applications of Solubility Curve Practice Problems With Answers in daily life. The third chapter will showcase real-world examples of how Solubility Curve Practice Problems With Answers can be effectively utilized in everyday scenarios.
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 6. In chapter 5, this book will draw a conclusion about Solubility Curve Practice Problems With Answers. This chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Solubility Curve Practice Problems With Answers.

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