

Understanding Solubility

What is the quantity of solute in a saturated solution at a certain temperature?

Activity 1

Prepare two solid solutes: a potassium salt and a copper sulphate. Heat 100 ml water and separate into two equal portions. Into each portion add a small amount of the potassium salt and stir thoroughly. Repeat the process with the copper sulphate. Observe the solubility of the solutes in water at room temperature. Now heat the water in each beaker and observe the solubility of the solutes in hot water. Record your observations. How does the solubility of the solutes change with temperature?

Discussion: The solubility of a solute in a solvent is the amount of solute that can dissolve in a given amount of solvent at a given temperature. The solubility of a solute in a solvent is affected by temperature.

Table 1: – Effect of Temperature

The following table shows the effect of temperature on the solubility of a solute in a solvent at different temperatures. The solubility of a solute in a solvent is the amount of solute that can dissolve in a given amount of solvent at a given temperature. The solubility of a solute in a solvent is affected by temperature.

Solute	Experiment 1 in 100 g water		Experiment 2 in 100 g water		Experiment 3 in 100 g water	
	Amount of solute dissolved (g)	Amount of solute dissolved (g)	Amount of solute dissolved (g)	Amount of solute dissolved (g)	Amount of solute dissolved (g)	Amount of solute dissolved (g)
1	10.0	10.0	10.0	10.0	10.0	10.0
2	10.0	10.0	10.0	10.0	10.0	10.0
3	10.0	10.0	10.0	10.0	10.0	10.0
4	10.0	10.0	10.0	10.0	10.0	10.0
5	10.0	10.0	10.0	10.0	10.0	10.0
6	10.0	10.0	10.0	10.0	10.0	10.0
7	10.0	10.0	10.0	10.0	10.0	10.0
8	10.0	10.0	10.0	10.0	10.0	10.0
9	10.0	10.0	10.0	10.0	10.0	10.0
10	10.0	10.0	10.0	10.0	10.0	10.0

1. Solubility of a solute in a solvent is the amount of solute that can dissolve in a given amount of solvent at a given temperature.

2. Solubility of a solute in a solvent is affected by temperature. The solubility of a solute in a solvent is higher at higher temperatures.

3. Solubility of a solute in a solvent is affected by temperature. The solubility of a solute in a solvent is higher at higher temperatures.

Solubility Pogil Answers

Joel Henry Hildebrand



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