Model 2 – Ternary Acids (Oxyacids)

Acid	Name of Acid in Aqueous Solution	Cation (+)	Polyatomic Anion	Polyatomic Anion Name
HCIO,	Chloric acid	H,O	C/O3-1	Chlorate
H,SO,	Sulfarous acid	24,00	503-2	Sulfise
H,SO,	Sulfuric acid	2 Hso*	504-3	Sulfate
H,PO,	Phosphorous acid	3 H,O-	P03-3	Phosphite
H,PO.	Phosphoric acid.	3 H50+	PO _a s-	Phosphate
HNO,	Nitric acid	H30+	NO."	Nitrate
HNO,	Nitrous acid	H.0*	NO ₂ t-	Nitrite
H,CO,	Carbonic acid	2H30+	CO,-2	Carbonase

- 6. Look at the formulas of the ternary acids in Model 2.
 - w. How are ternary acids different from binary acids in their structure?

Ternory acids contain three elements and they also contain the element oxygen

6. What number do you think the prefix "ter-" refers to?

"ter" refers to three

- 7. When ternary acids are mixed with water, ions will form. Fill in the table above with the formulas and names of the anions.
- 8. Examine the pairs of ternary acids in Model 2 that contain sulfur, phosphorus, and nitrogen. Each pair has one acid that ends in "-ic" and another that ends in "-ous." These endings are related to the name of the polyatomic anion found in the acid ("-ate" or "-ite"). Complete the statements below with the correct acid name ending.

Polyatomic anion ending is "-ate" → acid name ending is /C

Polyatomic anion ending is "-ite" → acid name ending is 045 ...

9. If the prefix "hydro-" were used to name a ternary acid, what problem would this create when

- naming HCIO? CIO3' has the name of Chlorate. "ate" must be replaced with "ic" If "hydro" is used as a prefix, it would be hydrochloric acid. This has the formula HCI, not HCIO3.

 The prefix "hydro" is not used in the name when an acid contain more than two elements and one of the elements is oxygen.

 "ato is replaced with it and "ite" is replaced with "ous".

Circle the acid(s) below that would be named beginning with the prefix "hydro-."

FESO.



H,CO.

HNO.

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Atherton Seidell

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Solutions and Solubilities Michael R. J.. Dack,1975

Solubilities of Inorganic and Organic Substances ,1907 Solubilities of Inorganic and Organic Substances ,1907 Solubilities of Inorganic and Organic Compounds Atherton Seidell, 1919 The Experimental Determination of Solubilities G. T. Hefter, R. P. T. Tomkins, 2003-08-01 In diesem Band werden vielf ltige experimentelle Verfahren zur Bestimmung von L slichkeiten beschrieben Commission on Solubility Data der IUPAC beschlo die Qualit t der L slichkeitsdaten international anzugleichen und zu verbessern indem verl liche Verfahren zur Ermittlung dieser Daten zusammengefa t ver ffentlicht werden alle Datentypen wurden hier ber cksichtigt ausf hrliche Angaben zur Zuverl ssigkeit der einzelnen Verfahren und zur Auswahl der jeweils geeigneten Methode **Solubilities of Inorganic and Organic Substances** Atherton Seidell,1919 Handbook of Aqueous Solubility Data Samuel H. Yalkowsky, Yan He, Parijat Jain, 2016-04-19 Over the years researchers have reported solubility data in the chemical pharmaceutical engineering and environmental literature for several thousand organic compounds Until the first publication of the Handbook of Aqueous Solubility Data this information had been scattered throughout numerous sources Now newly revised the second edition of The Solubility of Nonelectrolytes Joel Henry Hildebrand, Robert Lane Scott, 1950 Solubility Data Series . **Principles of Solution and Solubility** Kz Shinoda,1978 Solubilities of inorganic and organic substances Atherton Handbook of Aqueous Solubility Data Samuel H. Yalkowsky, 2003-03-26 Over the years researchers have Seidell,1911 reported solubility data in the chemical pharmaceutical engineering and environmental literature for several thousand organic compounds Until now this information has been scattered throughout the literature Containing over 16 000 solubility data points for more than 4 000 organic compounds Handbook of Aqueous **Solubility Data Series** International Union of Pure and Applied Chemistry. Commission of Equilibrium Data. Sub-Committee on Solubility Data, 1979 Solubilities of <u>Inorganic and Organic Compounds</u> Atherton Seidell, William F. Linke, 1952 First Outlines of a Dictionary of **Solubilities of Chemical Substances** Frank Humphreys Storer, 1864 Solubilities of Inorganic and Organic Compounds Atherton Seidell, 2018-10-16 This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it This work is in the public domain in the United States of America and possibly other nations Within the United States you may freely copy and distribute this work as no entity individual or corporate has a copyright on the body of the work Scholars believe and we concur that this work is important enough to be preserved reproduced and made generally available to the public To ensure a quality reading experience this work has been proofread and republished using a format that seamlessly blends the original graphical elements with text in an easy to read typeface We appreciate your support of the preservation process and thank you for being an important part of keeping this knowledge alive and relevant National Toxicology Program's Chemical Solubility Compendium Lawrence H. Keith, Douglas B.

Principles of Solution and Solubility Kozo Shinoda, 1991

Walters,1991-12-10 This book is a compilation of experimentally determined solubility ranges of over 1 700 compounds in the National Toxicology Program s Chemical Repository Each compound s solubility was determined in a consistent manner in one to six solvents Chosen were those most commonly used for toxicology studies spill cleanups and chemical synthesis or chemical reaction experiments These solvents include acetone 95% ethanol water dimethyl sulfoxide methanol and toluene Data for many of the research and industrial chemicals featured in this volume do not exist anywhere else If you are a toxicologist safety professional industrial hygienist or chemist this book is a valuable reference tool you ll find yourself using every day

Solubility Curves Charles Hertzel Frantz,1933

Solubility Joel Henry Hildebrand,1924

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