

Matching

Match each item with the correct statement below.

- | | |
|---------------------|----------------|
| A. solvation | E. electrolyte |
| B. weak electrolyte | F. colloid |
| C. aqueous solution | G. surfactant |
| D. solvent | |

- _____ 1. interferes with hydrogen bonding between water molecules
- _____ 2. dissolving medium
- _____ 3. homogeneous mixture of water and dissolved substances
- _____ 4. Solute ions or molecules are surrounded by solvent molecules.
- _____ 5. compound that will conduct current in the liquid state or in aqueous solution
- _____ 6. compound that ionizes incompletely in aqueous solution
- _____ 7. mixture in which particle size averages between 1 nm and 1000 nm

Match each item with the correct statement below.

- | | |
|----------------------|-------------------|
| A. dispersed phase | E. Tyndall effect |
| B. surface tension | F. suspension |
| C. Brownian motion | G. solute |
| D. dispersion medium | H. emulsion |

- _____ 8. inward force tending to minimize surface area of a liquid
- _____ 9. dissolved particle
- _____ 10. mixture in which particle size averages greater than 1000 nm in diameter
- _____ 11. Colloidal particles spread throughout a suspension.
- _____ 12. phenomenon observed when beam of light passes through a colloid
- _____ 13. chaotic movement of colloidal particles
- _____ 14. colloid of a liquid in a liquid

Multiple Choice

Identify the choice that best completes the statement or answers the question.

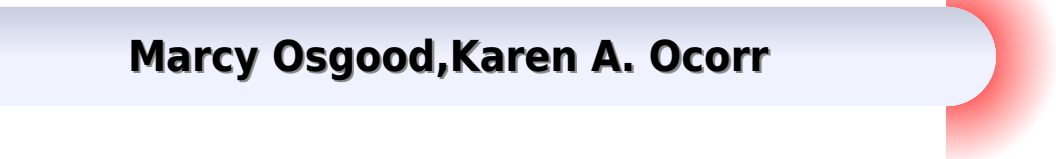
- _____ 15. How does the surface tension of water compare with the surface tensions of most other liquids?
 - A. It is lower.
 - B. It is about the same.
 - C. It is higher.
 - D. It is higher only when a surfactant is added.
- _____ 16. What causes water's low vapor pressure?

A. dispersion forces	C. hydrogen bonding
B. covalent bonding	D. ionic attractions
- _____ 17. What is the shape of the water molecule?

A. linear	B. tetrahedral	C. trigonal planar	D. bent
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- _____ 18. Which of the following is primarily responsible for holding water molecules together in the liquid state?

Water And Aqueous Systems Study Guide

Marcy Osgood, Karen A. Ocorr



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Study Guide for Atoms, Molecules, and Life Michael S. Matta, 1981 **The Absolute, Ultimate Guide to Lehninger Principles of Biochemistry** Marcy Osgood, Karen A. Ocorr, 2008-04-04 The Absolute Ultimate Guide combines an innovative study guide with a reliable solutions manual in one convenient printed volume *Water Quality Instructional Resources Information System (IRIS)*, 1979 Guide to the Analysis of Pesticide Residues Harry Phineas Burchfield, Donald E. Johnson, 1965 Ground Water Models National Research Council, Division on Engineering and Physical Sciences, Commission on Physical Sciences, Mathematics, and Applications, Committee on Ground Water Modeling Assessment, 1990-02-01 The discovery of toxic pollution at Love Canal brought ground water contamination to the forefront of public attention Since then ground water science and modeling have become increasingly important in evaluating contamination setting regulations and resolving liability issues in court A clearly written explanation of ground water processes and modeling Ground Water Models focuses on the practical aspects of model application It examines the role of models in regulation litigation and policy development explains ground water processes and describes specific applications for models presents emerging technologies and offers specific recommendations for better use of ground water science in policy formation **Research and Technology Program Digest** United States. National Aeronautics and Space Administration, **EPA-430/1**, 1979-05 *Guide to Graduate Study, Research, and Outreach in Water Resources and Related Fields at the University of Wisconsin-Madison* University of Wisconsin--Madison, 1991 Selected Water Resources Abstracts, 1990 **Indexes** United States. Environmental Protection Agency, 1983 **Study Questions and Problems in Inorganic Chemistry** Alexander Silverman, 1924 Student Study Guide to accompany Chemistry Martin Silberberg, Libby Weberg, 2005-01-06 **Thermodynamics and Kinetics of Water-Rock Interaction** Eric H. Oelkers, Jacques Schott, 2018-12-17 Volume 70 of Reviews in Mineralogy and Geochemistry represents an extensive review of the material presented by the invited speakers at a short course on Thermodynamics and Kinetics of Water Rock Interaction held prior to the 19th annual V M Goldschmidt Conference in Davos Switzerland June 19 21 2009 Contents Thermodynamic Databases for Water Rock Interaction Thermodynamics of Solid Solution Aqueous Solution Systems Mineral Replacement Reactions Thermodynamic Concepts in Modeling Sorption at the Mineral Water Interface Surface Complexation Modeling Mineral Fluid Equilibria at the Molecular Scale The Link Between Mineral Dissolution Precipitation Kinetics and Solution Chemistry Organics in Water Rock Interactions Mineral Precipitation Kinetics Towards an Integrated Model of Weathering Climate and Biospheric Processes Approaches to Modeling Weathered Regolith Fluid Rock Interaction A Reactive Transport Approach Geochemical Modeling of Reaction Paths and Geochemical Reaction Networks *Plant Physiology, 4th Edition* Pandey S.N. & Sinha B.K., 2009-11-01 Plant Physiology lucidly explains the operational mechanisms of plants based on up to date literature and with the help of numerous illustrations In addition to the theoretical aspects experiments have been

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Energy Research Abstracts, 1982

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Analysis of Nutrients, Selected Inorganic Constituents, and Trace Elements in Water from Illinois Community-supply Wells, 1984-91 Kelly L. Warner, 2000

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