

25. The number of σ bonds in $N \equiv N$ is
- 1
 - 2
 - 3
 - 4
26. The elements in an ionic compound are held together by
- electrostatic forces of attraction.
 - van der Waals forces
 - the spin of paired electrons.
 - the formation of hybrid orbitals.
 - an electron pair.
27. In every electrolytic and galvanic (voltaic) cell the anode is that electrode
- at which oxidation occurs.
 - which attracts cations.
 - at which electrons are supplied to the solution.
 - at which reduction occurs.
28. Metal X was plated from a solution containing cations of X. The passage of 48.25 C deposited 31 mg of X on the cathode. What is the mass of X (in grams) per mole of electrons?
- 47
 - 62
 - 93
 - 186
29. In a galvanic (voltaic) cell in which the reaction is $Cd + Cu^{2+} \rightarrow Cu + Cd^{2+}$ and the ions are at unit concentration (activity), the cell potential is
- $$\begin{array}{ll} Cd \rightarrow Cd^{2+} + 2e^- & 0.4021 \text{ V} \\ Cu \rightarrow Cu^{2+} + 2e^- & -0.344 \text{ V} \end{array}$$
- 0.1383 V
 - 0.4021 V
 - 0.344 V
 - 0.7461 V
 - 0.3677 V
30. In which reaction will an increase in total pressure at constant temperature favor formation of the products?
- $CaCO_3(s) \rightleftharpoons CaO(s) + CO_2(g)$
 - $H_2(g) + Cl_2(g) \rightleftharpoons 2HCl(g)$
 - $2NO(g) + O_2(g) \rightleftharpoons 2NO_2(g)$
 - $COCl_2(g) \rightleftharpoons CO(g) + Cl_2(g)$

Standard Potentials	E°
$Mg \rightarrow Mg^{2+} + 2e^-$	2.37V
$Al \rightarrow Al^{3+} + 3e^-$	1.66V
$Zn \rightarrow Zn^{2+} + 2e^-$	0.76V
$Fe \rightarrow Fe^{2+} + 2e^-$	0.44V
$Cu \rightarrow Cu^{2+} + 2e^-$	0.34V
$Ag \rightarrow Ag^+ + e^-$	-0.80V

31. Using only the metals Mg, Al, Zn, Fe, Cu and Ag, together with their 1 M salt solutions, a voltaic cell of the highest possible voltage would be constructed using electrodes of these metals.
- Mg and Ag
 - Mg and Fe
 - Zn and Cu
 - Al and Ag
 - Mg and Al
32. $E = E^\circ - 0.059/n \log Q$ (Nernst equation)
 $[H^+] = 1.0 \text{ M}$ initially, $P_{O_2} = 1.0 \text{ atm}$
- $$4e^- + O_2(g) + 4H^+(aq) \rightleftharpoons 2H_2O(l) \quad E^\circ = 1.23 \text{ V}$$
- Based on the information above, which statement is correct?
- $n = 1$, since one mole of oxygen is being considered.
 - Addition of base should result in an E value, which is less than 1.23 V.
 - E is independent of the pH of the solution.
 - $Q = \frac{[H_2O]^2}{[O_2][H^+]}$
33. The equilibrium constant for the gaseous reaction $C + D \rightleftharpoons E + 2F$ is 3.0 at 50 °C. In a 2.0 L flask at 50 °C are placed 1.0 mol of C, 1.0 mol of D, 1.0 mol of E, and 3.0 mol of F. Initially, the reaction will
- proceed at equal rates in both directions.
 - proceed more rapidly to form E and F.
 - proceed more rapidly to form C and D.
 - not occur in either direction.
- | Compound | ΔG_f° , kJ/mol |
|-----------|-----------------------------|
| $H_2O(l)$ | -237 |
| $H_2O(g)$ | -229 |
34. At 298 K the equilibrium constant for $H_2(g) + \frac{1}{2} O_2(g) \rightleftharpoons H_2O(l)$
- is larger than the K_{eq} for $H_2(g) + \frac{1}{2} O_2(g) \rightleftharpoons H_2O(g)$
 - will have a value of 1.0 at equilibrium.
 - cannot be computed since data on O_2 and

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ebook itbs compendium iowa test basic skills - Sep 24 2022

web the itbs test also known as the iowa test of basic skills is a nationally administered set of exams designed to assess the abilities of students subjects tested include language

[iowa assessments wikipedia](#) - Jun 02 2023

in the science section each question is designed to test a student s understanding of science students must choose the best answer for each question sample question choices see more

iowa tests of basic skills - Jan 29 2023

web overview of the iowa tests the iowa tests of basic skills are a series of norm referenced measures that assess student achievement in kindergarten through grade 8

iowa test of basic skills k12 academics - Mar 31 2023

web the iowa tests of basic skills itbs offer educators a diagnostic look at how their students are progressing in key academic areas

[iowa test of basic skills itbs practice test tests com](#) - Jul 03 2023

in the social studies section students are tested on their knowledge of social studies the social studies section is divided into 4 sections civics government see more

about the iowa tests of basic skills itbs funeducation - Dec 28 2022

web the iowa tests of basic skills itbs are standardized tests provided as a service to schools by the college of education of the university of iowa the tests are

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web the davidson institute has partnered with northwestern university s center for talent development ctd to offer a davidson specific above grade level version of the iowa test of basic skills for 3rd 5th graders this davidson specific testing opportunity is limited to the dates listed on the eligibility assessment page

iowa test of basic skills itbs davidson institute - Feb 15 2022

web the iowa test of basic skills exam itbs is a nationally standardized achievement test it is administered to students in grades k 12 itbs is available for homeschools and