

Solubility Product, K_{sp} - Worksheet 1

1) Write the dissociation equation and the solubility product (K_{sp}) expression for the dissociation of each of the following substances.

- | | | |
|---------------------------------|-----------------------------------|-------------------------------------|
| a) PbSO_4 (s) | c) $\text{Pb}(\text{NO}_3)_2$ (s) | e) Na_2S (s) |
| b) $\text{Mg}(\text{OH})_2$ (s) | d) AlCl_3 (s) | f) $\text{Ca}_3(\text{PO}_4)_2$ (s) |

2) Write the solubility product expression for the dissociation of each of the following solids in water.

- | | | |
|----------------------|-----------------------|----------------------|
| a) sodium chloride | c) silver sulfide | e) magnesium sulfite |
| b) calcium carbonate | d) aluminum hydroxide | f) barium phosphate |

3) The solubility of NiS is 1.0×10^{-12} mol/L. Calculate the value of K_{sp} for this compound.

4) The solubility of calcium hydroxide ($\text{Ca}(\text{OH})_2$) is 0.02 mol/L. Calculate the solubility product constant for this substance.

5) The solubility of silver sulfide (Ag_2S) is 1.3×10^{-6} mol/L. Calculate the K_{sp} value.

6) Experiments show that 0.0059 g of SrCO_3 will dissolve in 1.0 L of water. What is the K_{sp} for strontium carbonate?

7) The solubility of cobalt (II) sulfide is 3.8×10^{-3} g/L. Calculate the value of K_{sp} .

Solubility Product, K_{sp} - Worksheet 2

1. Calculate the solubility (in mol/L) for each of the following compounds:

- | | |
|-----------------------------|--------------------------------|
| a. FeS | $K_{sp} = 3.7 \times 10^{-19}$ |
| b. $\text{Mg}(\text{OH})_2$ | $K_{sp} = 9.0 \times 10^{-12}$ |
| c. Ag_2S | $K_{sp} = 1.6 \times 10^{-49}$ |
| d. AgCN | $K_{sp} = 2.0 \times 10^{-12}$ |

2. What is the solubility of silver iodide in grams per litre? ($K_{sp} = 8.3 \times 10^{-17}$)

3. The K_{sp} value for BaSO_4 is 1.5×10^{-9} . Calculate the number of moles of this salt that will dissolve in 1000 mL of water.

4. Calculate the mass of CuS that will dissolve in 1.0 litre of water. $K_{sp} = 4.0 \times 10^{-38}$

5. Calculate the mass of barium carbonate that will dissolve in 100 mL of water. The K_{sp} of BaCO_3 is 8.1×10^{-9} .

Solubility Product Pogil Answers

Sherry Lewin

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Solubility Product Pogil Answers:

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Solubility Product Pogil Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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Table of Contents Solubility Product Pogil Answers

1. Understanding the eBook Solubility Product Pogil Answers
 - The Rise of Digital Reading Solubility Product Pogil Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Solubility Product Pogil Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Platform
 - User-Friendly Interface
4. Exploring eBook Recommendations from Solubility Product Pogil Answers
 - Personalized Recommendations
 - eBook Platform User Reviews and Ratings
 - eBook Platform Bestseller Lists
5. Accessing Solubility Product Pogil Answers Free and Paid eBooks

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 - Solubility Product Pogil Answers Compatibility with Devices
 - Solubility Product Pogil Answers Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Solubility Product Pogil Answers
 - Highlighting and Note-Taking Solubility Product Pogil Answers
 - Interactive Elements Solubility Product Pogil Answers
 8. Staying Engaged with Solubility Product Pogil Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Solubility Product Pogil Answers
 9. Balancing eBooks and Physical Books Solubility Product Pogil Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Solubility Product Pogil Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Solubility Product Pogil Answers
 - Setting Reading Goals Solubility Product Pogil Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Solubility Product Pogil Answers
 - Fact-Checking eBook Content of Solubility Product Pogil Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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