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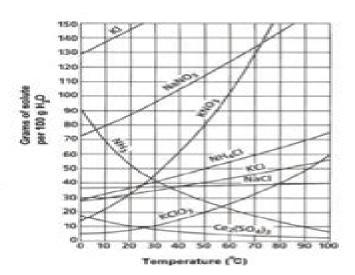
Write the correct term (or terms) in the space provided.

- 1. As temperature increases, the solubility of gases in liquids generally
- 2. When a solute dissolves and recrystallizes at the same rate, the solution is at
- 3. A substance that does not dissolve in a polar solvent is probably
- 4. To carry an electric current, a solution must contain
- 5. A solution that contains more dissolved solute than a saturated solution contains under the same conditions is called a(n)
- 6. The solubility of CO<sub>2</sub>(g) would with increasing solvent temperature.

Based on the graph answer the following questions:

- Which compound is LEAST soluble at 10 °C?
- 2. How many grams of KCl can be dissolved in 100g of water at 80°C?
- 3. How many grams of NaCl can be dissolved in
- 100g of water at 90°C?

  4. At 40°C, how much KNO3 can be dissolved in 100g of water? \_\_\_\_\_
- 5. A saturated solution of KClO3 is formed from one hundred grams of water. If the saturated solution is cooled from 90°C to 50°C, how many grams of precipitate are formed?



What are 3 factors that affect solubility?

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**Ian Pickup** 

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