#### CE5504 - Surface Water Quality Modeling

#### A Toxic Substance Model for a Well-Mixed Lake

To this point, we have dealt with what are often termed conventional pollutants, e.g. nitrogen, phosphorus, organic carbon, etc. Now our attention will move to toxic substances, substances which differ from conventional pollutants in several ways important to the modeler:

- natural versus alien: problems relating to conventional pollutants typically involve the natural cycle of organic production and decomposition, most often over stimulation of natural processes. Toxic substances do not occur naturally and the problem is one of interference with natural processes.
- aesthetics versus health: a case could be made that mitigation of problems relating to conventional pollutants largely seeks to eliminate aesthetic problems. Although such a case would be overstated, it is clear that management of toxic substances focuses on health concerns, e.g. drinking water and aquatic food stuffs.
- few versus many: conventional water quality management deals with on the order of ten 'pollutants'. In contrast, there are tens of thousands of organic chemicals that may be introduced to the environment.
- single versus multiple phase: conventional pollutants are largely modeled as single species, although some partitioning is made between the dissolved and particulate phases. Such partitioning is critical for modeling toxic substances, because they tend to exist in gaseous, dissolved and particulate phases.

It is the liquid/solid partitioning which most significantly impacts the manner in which toxic substances are modeled. A mechanistic treatment of partitioning is important because key source/sink processes act selectively on either the dissolved or particulate form. For example, volatilization acts solely on the dissolved form, settling only on the particulate form, and decay reactions on both.

Mathematically, the total contaminant concentration (mg-m<sup>-3</sup>) is given by,

$$c = c_d + c_p$$

where  $c_d$  is the dissolved component and  $c_p$  is the particulate component, each representing a fixed fraction of the total concentration,

$$c_d = F_d \cdot c$$

and

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**Michael Rajnik** 

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