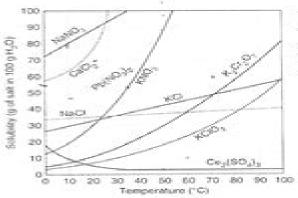
Worksheet: Solubility Graphs

Use the provided solubility graph to answer the following questions:



For questions 1 - 4 an amount of solute is given, and a temperature is stated. If all of the solute could be dissolved in 100 g of water at the given temperature, would the resulting solution be unsaturated, saturated, or supersaturated?

60 g KCl at 70 °C

Super Sutvented

2. 10 g KClO3 at 60 °C Us Saturated

3. 80 g NaNO 3 at 10 °C Saturated

4. 70 g Cacle at 20 °C Unsaturated

For questions 5 - 8 a solute and temperature are given. Tell how many grams of each solute must be added to 100 g of water to form a saturated solution at the given temperature.

5. Pb(NO₃)₂ at 10 °C

7 NaCl at 20 °C

6. Ce2(SO4)3 at 50 °C

8. K2Cr2O2 at 50 °C

For questions 9 and 10 underline the solution that is more concentrated.

- At 10 °C: a saturated solution of KNO₃ or a saturated solution of CaCl₂?
- 10. At 50 °C: a saturated solution of KNO3 or an unsaturated solution of NaNO3 consisting of 90 g of the solute dissolved in 100 g of water.

For questions 11 - 12, show your work and circle your final answer.

11. If 115 g KNO3 are added to 100 g of water at 35 °C, how many grams do not dissolve?

will not dissolve.

12. What mass of KCl would be needed to form a saturated solution if the KCl was dissolved in 200 g of water at 80 °C?

> CHEMISTRY: A Study of Matter 10.12

Solubility Temperature Graphs Chapter 14 Answers

Ralph H. Petrucci, William S. Harwood

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