



# Solubility Study Guide

Name \_\_\_\_\_

Teacher: \_\_\_\_\_

Date: \_\_\_\_\_

Period \_\_\_\_\_

## References

Chemistry in the Community pp 45-65  
Water properties concept map  
Solubility concept map  
Solubility, graphing and analysis  
Solubility Practice: A Curvy Subject  
Study Solvent Lab

## Vocabulary

Understand and apply the definitions or symbols of the following terms.

Colligative properties  
Distillation  
Filtration  
Homogeneous  
Mixtures  
Nonpolar  
Polar  
Pure substances  
Solubility

Solute  
Solvent  
Solution  
Supersaturated  
Saturated  
Unsaturated  
 $\delta^+$  and  $\delta^-$

## Targeted Skills

1. Describe the structure of water and relate its unusual physical properties to its structure, high boiling point, surface tension, high specific heat.
2. Draw the structure of water and relate its structure to the ability of water to act as a universal solvent.
3. Compare/contrast the solubilities of various ionic and covalent solutes.
4. Apply: Like dissolves like:  
Polar solvents dissolve Polar and Ionic solutes  
Nonpolar solvents dissolve Nonpolar solutes
5. Describe factors that affect the rate of dissolving, such as particle size, stirring, temperature, nature of the solid solutes.
6. Interpret information from a solubility curve (graph).

## Make sure you remember from past tests?

Atomic structure  
Significant figures  
Accuracy & precision

## Equations: (extension)

Solution concentration =  $\frac{\text{amount of solute}}{\text{amount solute} + \text{solvent}}$

## Test Breakdown

30 Multiple-Choice Problems (max.) – 60%

- 1 Free Response Problem chosen from the Free Response Test Bank – 10%
- 1 Open-Ended Problem chosen from the Open-Ended Test Bank – 10%
- 1 Performance Assessment – 20%

# Solubility Study Guide

**Andrew Roberts**



## Solubility Study Guide:

Chemical Principles Study Guide/Solutions Manual John Krenos, Joseph Potenza, Carl Hoeger, 2007-01-18 Written for general chemistry courses Chemical Principles helps students develop chemical insight by showing the connection between chemical principles and their applications

**Chemical Principles Student's Study Guide & Solutions Manual** John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better The solutions manual section includes detailed answers and explanations to the odd numbered exercises in the text

*The Practice of Chemistry Study Guide & Solutions Manual* Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

**ACT Prep Plus 2024: Study Guide includes 5 Full Length Practice Tests, 100s of Practice Questions, and 1 Year Access to Online Quizzes and Video Instruction** Kaplan Test Prep, 2023-07-04 Kaplan is an Official Teaching Partner of the ACT Kaplan's ACT Prep Plus 2024 has the detailed subject review practice tests and expert strategies you need to be prepared for test day This ACT prep book includes hundreds of practice questions online practice tests and video lessons from our experts to help you face test day with confidence We're so certain that ACT Prep Plus offers the guidance you need that we guarantee it After studying with our online resources and book you'll score higher on the ACT or you'll get your money back Essential Review 5 full length Kaplan practice tests with detailed answer explanations 1 printed in the book and 4 tests online One year access to our online center with additional Qbank and videos to help guide your study Pre quizzes to help you figure out what you already know and what you can skip Mixed practice quizzes after every chapter to assess how much you've learned A practice question at the beginning of each lesson to help you quickly identify its focus and dedicated practice questions after every lesson to test your comprehension Efficient Strategy On Test Day strategy notes in every math chapter to help you remember that the ACT math test is primarily a strategy test Reflect pages that help you evaluate your comfort level with the topics and make a plan for improving before the test after completing each chapter Online study planning tool helps you target your prep no matter how much time you have before the test Expert Guidance We know the test Our learning engineers have put tens of thousands of hours into studying the ACT and we use real data to design the most effective strategies and study plans Kaplan's books and practice questions are written by veteran teachers who know students every explanation is written to help you learn We invented test prep Kaplan kaptest.com has been helping students for over 80 years Trying to figure out your college plan Kaplan's KapAdvisor is a free college admissions planning tool that combines Kaplan's expertise with the power of AI

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end

of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

**Study Guide for Chemical Principles [by] Steven S. Zumdahl** Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples

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**Safety Professional's Reference and Study Guide** W. David Yates, 2010-12-09 While journals provide up to the minute information and textbooks prepare you for certification maintaining a library suitable for a comprehensive practice can become quite costly A collection of information drawn from textbooks journals and the author's 25 years of experience Safety Professional's Reference and Study Guide provides as the ti

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**Hazmat Chemistry Study Guide (Second Edition)** Jill Meryl Levy, 2011

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**Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition** John McMurry, Susan McMurry, 1999

Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self tests at the end of each chapter

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