

Name : _____ Date : _____

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Solubility Rules Practice Worksheet

1. Classify each substance as being soluble or insoluble in water.

I. $\text{Mg}(\text{PO}_4)_2$ - _____

VI. KBr - _____

II. KOH - _____

VII. $\text{Pb}(\text{CO}_3)_2$ - _____

III. NiCl_2 - _____

VIII. PbI_2 - _____

IV. NH_4OH - _____

IX. BaSO_4 - _____

V. Hg_2SO_4 - _____

X. NiCl_2 - _____

2. Show the ions that formed the following compounds:

I. $\text{Zn}_3(\text{PO}_4)_2$

II. Al_2S_3

III. Iron (III) sulfide

IV. Ammonium cyanide

3. Form 4 water-soluble compounds by combining ions from the ions below:

Cl^-

CO_3^{2-}

PO_4^{3-}

Li^+

Sr^{2+}

4. Identify the precipitate in the following reaction. Circle the correct answer.



5. Name or give the chemical formula for each of the following compounds and state whether they are soluble (will dissolve) or insoluble (will not dissolve) in solution.

Chemical Formula	Name	Solubility
$\text{Zn}_3(\text{PO}_4)_2$		
	Silver bromide	
KNO_3		
	Aluminum sulfide	
	Silver acetate	

Solubility Practice Problems

Paul A Krieger

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Solubility Practice Problems:

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Comprehensive Chemistry XI Dr. B. Kapila, S. K. Khanna, 2010-11 Comprehensive chemistry according to the new syllabus prescribed by Central Board of Secondary Education CBSE **Chemistry Calculations for Beginners** John Obimakinde, Samuel Obimakinde, Ebenezer Obimakinde, Fredrick Akinbolade, 2025-05-30 With decades of combined experience as science teachers at both school and undergraduate levels the authors have recognised that one of the greatest challenges faced by students studying chemistry is grasping the complexity of the numerous numerical problems found in most parts of the subject This text is crafted to provide a clear and accessible pathway to overcoming this challenge by assisting students especially novices or those with minimal knowledge of the subject in performing chemistry calculations The content covers fundamental calculations crucial to understanding the principles of chemistry making it an invaluable tool for students aiming to excel in their studies Key features Designed with a student friendly approach including detailed explanation of chemical concepts underlying each type of calculation step by step explanations alternative methods for solving problems numerous practice exercises answers to practice exercises and appendices The book is tailored to suit various curricula ensuring relevance for a diverse audience Encompasses a wide range of calculations offering students a thorough understanding of essential chemistry concepts Serves as an excellent resource for exam preparation and equips students with skills applicable to future scientific endeavours Employs straightforward language to ensure ease of understanding for beginners Uses IUPAC conventions underscoring the universal nature of chemistry **Introductory Basics Of Chemistry** Dr J. Sai Chandra, 2022-05-07 Our Chemistry Reference Book adheres to the scope and sequence of most general chemistry courses nationwide We strive to make chemistry as a discipline interesting and accessible to students With this objective in mind the content of this Reference Book has been developed and arranged to provide a logical progression

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