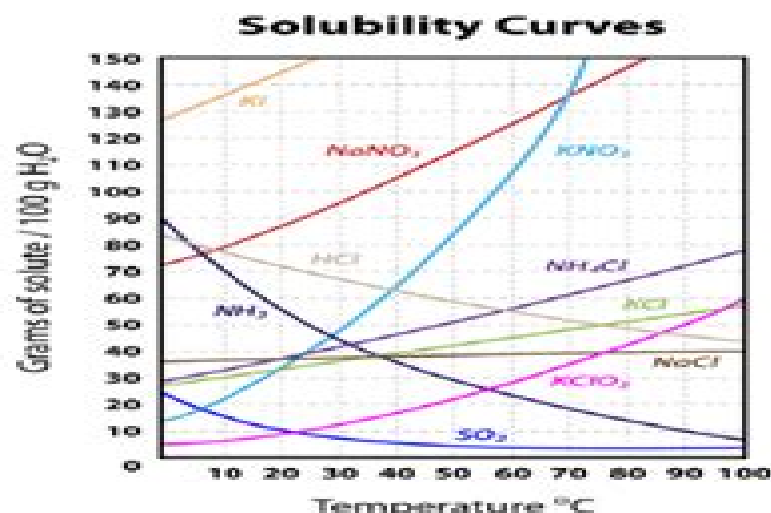


WS 6.1: Interpreting Solubility Curves

Directions: Use the Solubility Curve Table below to answer all questions. So you do not have to keep flipping your paper over you might want to put the solubility curve on your phone or computer screen when answering the questions on the back. All questions assume the amount of water is 100 ml.



- What is the solvent in this Solubility Curve?
water
- Saturation points are given per how much solvent?
100 ml or 100 g
- What is the saturation point of NaNO_3 at 10°C , 40°C , & 80°C ?
@ 10°C = ~79 g
@ 40°C = ~105 g
@ 80°C = ~146 g
- How much NH_3 can you dissolve in water at 10°C , 30°C , & 90°C ?
@ 10°C = ~70 g
@ 30°C = ~45 g
@ 90°C = ~10 g
- What is the maximum amount of KCl that can be dissolved in water at 20°C , 50°C , & 90°C ?
@ 20°C = ~33 g
@ 50°C = ~42 g
@ 90°C = ~53 g
- If you want to make a saturated solution of NaCl how many grams would you dissolve in 100 grams of water @ 80°C ?
About 40 grams
- Except for KCl What substance is the most soluble in water at 20°C ?
 NaNO_3
- Except for KCl what substance is the most soluble in water at 72°C ?
 KNO_3
- Look up The boiling points for SO_2 , HCl , and NH_3 and write them below:
B.P. SO_2 = -10°C
B.P. HCl = -85°C
B.P. NH_3 = -33°C
- Based on the boiling points you looked up, are SO_2 , HCl , & NH_3 solids, liquids, or gases at 0°C ? (and anywhere else on the solubility curve)
gases

Solubility Curves Answers

G Orfield

A red circular graphic with a gradient, appearing as a stylized arrow or a decorative element, located to the right of the author's name.

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- Chapter 2: Essential Elements of Solubility Curves Answers
- Chapter 3: Solubility Curves Answers in Everyday Life
- Chapter 4: Solubility Curves Answers in Specific Contexts
- Chapter 5: Conclusion

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5. In chapter 4, the author will scrutinize the relevance of Solubility Curves Answers in specific contexts. The fourth chapter will explore how Solubility Curves Answers is applied in specialized fields, such as education, business, and technology.
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The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Solubility Curves Answers.

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web electrostatics 2 1 the electrostatic field to calculate the force exerted by some electric charges $q_1 q_2 q_3$ the source charges on another charge q the test charge we can use the principle of superposition this principle states that the

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web answer d somewhere in region d justification with the test charge and c 1 being negative there is a repulsive force on the test charge to the right from c 2 there is an attractive force on the test charge to the left by referring to coulomb s law $\square\square\square\square$ 1 $\square\square$ 2 $\square\square$ 2 we know that the force

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