The number of moles of substances, when compared and reduced to the lowest terms, yields a mole ratio, as shown in the sample below. Indicate the mole ratio for each of the following equations:

$$4H_{2} + 2O_{2} \rightarrow 4H_{2}O$$

$$2: 1: 2$$

$$H_{2} + S \rightarrow H_{2}S$$

$$4AI + 3O_{2} \rightarrow 2 AI_{2}O_{3}$$

$$37n + 6t ICI \rightarrow 37nCI_{2} + 3H_{2}$$

$$SO_{2} + \frac{1}{2}O_{2} \rightarrow SO_{3}$$

$$2: 1: 2$$

$$2: 1: 1$$

$$37n + 6t ICI \rightarrow 37nCI_{2} + 3H_{2}$$

$$37n + 6t ICI \rightarrow 37nCI_{2} + 3H_{2}$$

$$2: 1: 1$$

$$2: 1: 1$$

Convert the following masses to moles by using the information on your periodic table.

a)
$$34.0 \text{ g NH}_3$$
 $FW = 17.09 \text{ kmol}$
 34.0 g NH_3 $FW = 17.09 \text{ kmol}$
 2.00 mol

b) 51.4 g Na_2O $FW = 62.09 \text{ kmol}$
 $51.4 \text{ g } \left(\frac{1 \text{ mol}}{62.09}\right) = 0.829 \text{ mol}$

c) 4.50 g H_2SO_4 $FW = 98.1 \text{ g kmol}$
 $4.50 \text{ g } \left(\frac{1 \text{ mol}}{98.1 \text{ g}}\right) = 0.0459 \text{ mol}$

d) 229 g Ca(NO_3)_2 $FW = 164.1 \text{ g kmol}$
 229 g Ca(NO_3)_2 $FW = 164.1 \text{ g kmol}$
 229 g Ca(NO_3)_2 $FW = 164.1 \text{ g kmol}$
 36.0 g NH_4OH $FW = 35.0 \text{ g kmol}$
 36.0 g NH_4OH $FW = 35.0 \text{ g kmol}$
 36.0 g NH_4OH $FW = 35.0 \text{ g kmol}$
 36.0 g NH_4OH $FW = 35.0 \text{ g kmol}$
 36.0 g NH_4OH $FW = 35.0 \text{ g kmol}$

Convert the following to grams.

a)
$$17.1 \text{ moles H}_2O = F\omega = 18.00 \text{ mol}$$
 308 g. 308 g .

Skills Practice 34 Chemistry Worksheet Key

Annelies Wilder-Smith

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3 d x 5 e x 5 note students may have various strategies for solving 3x 5 20 such as using fact families 3x 5 20 so 3x 20 5 if 3x 15 then x 15 3 using an undoing metaphor begin with 3x 5 20 subtract 5 from each side and then divide by 3 on each side using properties of equality

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a jose 15 3 5 mph mario 21 3 7 mph melanie 27 3 9 mph b jose 7 5 35 mi mario 7 7 49 mi melanie 7 9 63 mi c cycling trip d jose about 33 mi mario about 46 mi melanie about 59 mi e jose 14 hours mario 10 hours melanie about 7 75 hours f the faster the cyclist the steeper the graph g

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