# **Electron Configuration Practice Worksheet**

In the space below, write the unabbreviated electron configurations of the following elements:

2) iron 15° 25° 25° 35° 45° 36°

3) bromine \\\^2 \frac{2}{2} \frac{6}{3} \frac{5}{4} \frac{6}{3} \frac{6}{4} \frac{1}{3} \frac{1}{4} \frac{6}{3} \frac{1}{4} \frac{1}{3} \frac{1}{4} \

5) neptunium 1525253535453575455546 654656575 5F4661

In the space below, write the short hand electron configurations of the following elements:

6) cobalt [A] 45 31

7) silver [kr] 5. 4d"

8) tellurium \_ [kc] 5c 4d 5p

9) radium (R-) 751

10) lawrencium (C.) 75" 55" 6d"

Determine what elements are denoted by the following electron configurations:

13) [Kr] 5s24d105p3 \_\_\_\_\_\_\_

14) [Xe] 6s<sup>2</sup>4f<sup>14</sup>5d<sup>6</sup>

15) [Rn] 7s25f11 \_\_\_\_\_\_\_

Determine which of the following electron configurations are not valid:

16) 1s<sup>2</sup>2s<sup>2</sup>2p<sup>6</sup>3s<sup>2</sup>3p<sup>6</sup>4s<sup>2</sup>4d<sup>19</sup>4p<sup>5</sup>\_\_\_\_\_

17) 1s<sup>2</sup>2s<sup>2</sup>2p<sup>6</sup>3s<sup>2</sup>3d<sup>5</sup>

18) [Ra] 7s<sup>2</sup>5f<sup>8</sup>

20) [Xe] \_ (-- + be one (-- 1 -- 1 -- 1 -- 1

# **Skill Practice 13 Configuration Practice Answers**

Jeff Gold, Rick Holden, Paul Iles

#### **Skill Practice 13 Configuration Practice Answers:**

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