

Worksheet 13 – Intermolecular Forces

Chemical bonds are **intramolecular** forces, which hold atoms together as molecules.

Intermolecular forces exist **between** atoms and molecules, determining their **physical** properties; solubility, boiling points, melting points.

Intermolecular forces (IMF) can be ranked qualitatively using Coulomb's Law:

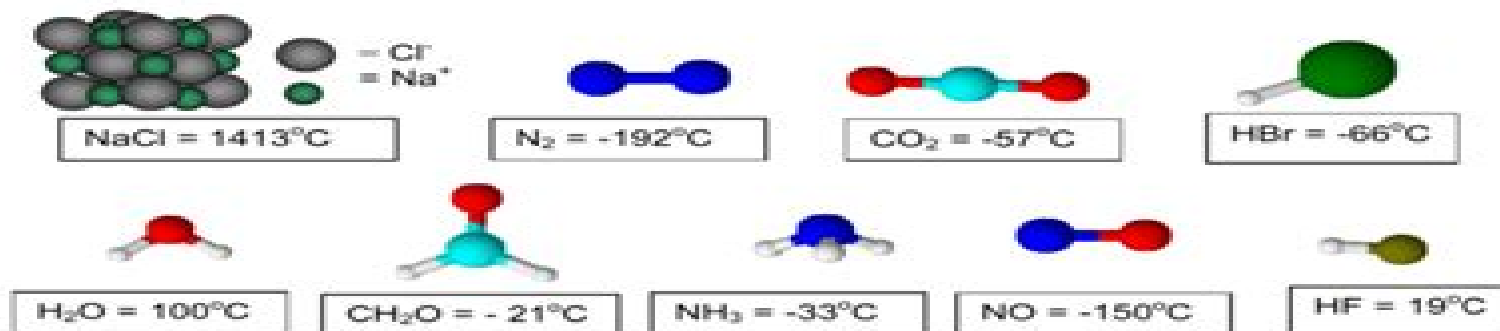
$$\text{force} \propto \frac{Q_1 Q_2}{r}$$

where Q_1 and Q_2 are charges and r is the distance between the charges.

The consequences are:

- the bigger the charges, the stronger the interaction
- the closer the charges, the stronger the interaction

Shown below are the structures and boiling points of nine chemical species. Use this information to answer the following questions.



1. Which species has the **highest** boiling point? _____
This is indicative of the **strongest IMF**.

- a. How does it differ from all of the other compounds?
- b. Look at Coulomb's law and identify each variable in the equation to the components of this species.
 Q_1 (positive charge) = _____ Q_2 (negative charge) = _____

This is called an _____ interaction (full positive and negative charges).

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Holt Chemistry Holt Rinehart & Winston, 2003-01-24

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In a digitally-driven earth where screens reign supreme and instant connection drowns out the subtleties of language, the profound techniques and mental nuances concealed within phrases frequently go unheard. However, nestled within the pages of **Skills Worksheet Concept Review Section Intermolecular Forces Answers** a interesting literary prize pulsing with organic thoughts, lies an extraordinary quest waiting to be undertaken. Written by a talented wordsmith, that charming opus invites viewers on an introspective journey, softly unraveling the veiled truths and profound influence resonating within the fabric of each word. Within the psychological depths of this poignant evaluation, we can embark upon a honest exploration of the book is primary styles, dissect their captivating publishing model, and yield to the strong resonance it evokes heavy within the recesses of readers hearts.

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