

WORKSHEET ON SOLUTION CONCENTRATIONS**INTRODUCTION**

A solution is made up of two parts, the **SOLVENT** and the **SOLUTE**. The solution could be **CONCENTRATED** or **DILUTE**.

SOLVENT

The liquid which dissolves the solute. It is usually water.

SOLUTE

The substance which dissolves - often a solid, but not always.

CONCENTRATED

There is a lot of solute and a little water.

DILUTE

There is a lot of water and only a little solute.

When we measure out 50cm^3 of a solution, e.g., hydrochloric acid, we measure out some solvent (water) and some acid together. In any reactions of the acid, it is the HCl particles which are involved. The water is just a carrier for the acid, so when we measure out a volume of the solution, we want to know how much acid it contains. To do this, we need to define the **CONCENTRATION** of the solution. This tells us *how much* solute there is in a *fixed volume* of the solution. In chemistry, "*how much*" means **MOLES** and the "*fixed volume*" is taken as 1 cubic decimetre (1dm^3).

[NB: 1dm^3 is 1000cm^3 and is the same volume as 1 litre.]

DEFINITION

Concentration of solution (moles per dm^3)	=	<u>number of moles of solute</u> <u>volume of solution in dm^3</u>
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- You must learn this relationship and be able to use it and rearrange it to calculate moles of solute or volume of solution, given the other two quantities.
- Remember that the volume must be in dm^3 , so convert from cm^3 if necessary.
- The units for concentration are *moles per cubic decimetre* which is written mol.dm^{-3}
- Lab bottles use the letter *M* (meaning *molar*) and this is the same thing as mol.dm^{-3}

RELATIONSHIPS

Use the triangle opposite to help you answer the questions which follow. Some worked examples are given to help you.



Examples:-

- 1) What is the concentration of a solution which contains 2 moles of copper (II) sulphate in 4dm^3 of solution?

Ans: concentration = moles ÷ volume = $2 \div 4 = \underline{0.5 \text{ mol.dm}^{-3}}$

- 2) How many moles of sodium hydroxide are there in 200cm^3 of a solution of concentration 0.2 mol.dm^{-3} ? (Remember that $200\text{cm}^3 = 200 \div 1000 = 0.2\text{dm}^3$)

Ans: moles = concentration × volume = $0.2 \times 0.2 = \underline{0.04 \text{ moles}}$

Skill Practice 39 Concentration Practice Answers

Stephen Lamb, Russell W. Rumberger



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Exercise Physiology William J. Kraemer, Steven J. Fleck, Michael R. Deschenes, 2011-03-01 Designed for undergraduate course work this exercise physiology textbook unites research and theory with real world application so students can easily relate to the concepts being presented The unique applied approach fully engages you in discovering how the human body works and responds to exercise You ll not only gain a solid foundation in exercise physiology concepts you ll also learn how to apply these concepts on the job to optimize athletic performance and well being Moreover you ll come to understand the vital health benefits of exercise and physical activity for all individuals at all ages including special populations Beginning with basic exercise physiology concepts the text progressively builds your knowledge by integrating these concepts into practical discussions of nutrition and training The text stresses a research based approach enabling you to locate and evaluate the evidence you need to make good decisions Numerous examples further underscore the importance of basic concepts and research in addressing real life challenges in exercise and athletic training

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Molecular Exercise Physiology Henning Wackerhage, 2014-02-24 Molecular Exercise Physiology An Introduction is the first student friendly textbook to be published on this key topic in contemporary sport and exercise science It introduces sport and exercise genetics and the molecular mechanisms by which exercise causes adaptation The text is linked to real life

sport and exercise science situations such as what makes people good at distance running what DNA sequence variations code for a high muscle mass or by what mechanisms does exercise improve type2 diabetes The book includes a full range of useful features such as summaries definitions of key terms guides to further reading review questions personal comments by molecular exercise pioneers Booth Bouchard and leading research in the field as well as descriptions of research methods A companion website offers interactive and downloadable resources for both student and lecturers Structured around central themes in sport and exercise science such as nutrition endurance training resistance training exercise chronic disease and ageing this book is the perfect foundation around which to build a complete upper level undergraduate or postgraduate course on molecular exercise physiology [Slowpitch Softball Hitting Mini eBook](#) , *BITES: A Compilation of Scholarly Writings* Dr. Susan Ramilo-Butac,2023-09-08 BITES A Compilation of Scholarly Writings by Dr Susan Ramilo Butac

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The Effective Primary School Classroom Joan Dean, 2005 This accessible book provides student teachers and new teachers with a good overview of how to organise the busy primary classroom **Inequality in Key Skills of City Youth** Stephen Lamb, Russell W. Rumberger, 2024-01-01 This groundbreaking research volume addresses the topic of educational inequality from a global perspective It includes 16 chapters from an international group of scholars who examine how well city school systems from around the world are preparing young people particularly poor and minority students with the skills they will need for further study work and life overall While skills in key domains such as science math language and civics have been center stage in international comparisons there has been growing recognition of the effects that education has on the development of broader sets of capabilities such as social and emotional skills also known as noncognitive or 21st century skills that can affect the success of students in school and beyond This volume aims to address the shortage of international data on the wide range of skills that students need to learn enabling researchers to compare the types and causes of

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