

Chapter 10 Problems

$$\begin{aligned} 10\text{-}2.1 \quad C_P &= C_1 R_3/R_1 = 0.1 \mu\text{F} \times 1 \text{ k}\Omega/547 \Omega = 0.183 \mu\text{F} \\ R_P &= R_1 R_4/R_3 = 547 \Omega \times 666 \Omega/1 \text{ k}\Omega = 364.3 \Omega \end{aligned}$$

$$\begin{aligned} D &= 1/(\omega C_P R_P) \\ &= 1/(2 \pi \times 1 \text{ kHz} \times 0.183 \mu\text{F} \times 364.3 \Omega) = 2.39 \end{aligned}$$

10-3.1 From Ex. 10-2,

$$\begin{aligned} C_4 &= C_5 = 0.068 \mu\text{F} \text{ \& } R_4 = R_5 = 183.8 \Omega \\ R_3 &= C_4 R_4/C_1 = 0.068 \mu\text{F} \times 10 \text{ k}\Omega/0.1 \mu\text{F} = 6.8 \text{ k}\Omega \\ C_3 &= R_4 C_1/R_4 = 183.8 \Omega \times 0.1 \mu\text{F}/6.8 \text{ k}\Omega = 2700 \text{ pF} \end{aligned}$$

$$\begin{aligned} 10\text{-}4.1 \quad L_{S(\min)} &= C_3 R_1 R_{4(\min)} \\ &= 0.1 \mu\text{F} \times 100 \Omega \times 100 \Omega = 1 \text{ mH} \\ L_{S(\max)} &= C_3 R_1 R_{4(\max)} = 0.1 \mu\text{F} \times 100 \Omega \times 1 \text{ k}\Omega = 10 \text{ mH} \\ Q_{(\max)} &= 2 \pi f L_{S(\max)}/R_{3(\min)} \\ &= 2 \pi \times 10 \text{ kHz} \times 10 \text{ mH}/100 \Omega = 6.28 \\ Q_{(\min)} &= 2 \pi f L_{S(\min)}/R_{3(\max)} \\ &= 2 \pi \times 10 \text{ kHz} \times 1 \text{ mH}/1 \text{ k}\Omega = 0.0628 \end{aligned}$$

$$\begin{aligned} 10\text{-}5.1 \quad R_5 &= R_1 = 200 \Omega \\ \text{Coil resistance} &= R_5 - 185 \Omega = 15 \Omega \\ L_5 &= L + 2 M = 44 \text{ mH} + (2 \times 20 \text{ mH}) = 84 \text{ mH} \end{aligned}$$

$$\begin{aligned} 10\text{-}6.1 \quad C_x + C_P &= R_3 C_1/R_4 \\ &= 10.29 \text{ k}\Omega \times 0.1 \mu\text{F}/14.7 \text{ k}\Omega = 0.07 \mu\text{F} \\ C_x &= (C_x + C_P) - C_P = 0.07 \mu\text{F} - 0.068 \mu\text{F} = 2000 \text{ pF} \\ R_P \parallel R_x &= R_1 R_4/R_3 \\ &= 308.9 \text{ k}\Omega \times 14.7 \text{ k}\Omega/10.29 \text{ k}\Omega = 442.57 \text{ k}\Omega \\ R_P \parallel R_x &= 1/(1/R_P + 1/R_x) \text{ or } 1/(R_P \parallel R_x) = (1/R_P + 1/R_x) \\ \text{Giving, } 1/(442.57 \text{ k}\Omega) &= (1/553.1 \text{ k}\Omega) + (1/R_x) \\ \text{So, } R_x &= 1/[(1/442.57 \text{ k}\Omega) - (1/553.1 \text{ k}\Omega)] = 2.2 \text{ M}\Omega \end{aligned}$$

$$\begin{aligned} 10\text{-}1 \quad C_{3(\max)} &= C_1 R_{3(\max)}/R_{4(\min)} \\ &= 0.1 \mu\text{F} \times 200 \text{ k}\Omega/1 \text{ k}\Omega = 20 \mu\text{F} \\ C_{3(\min)} &= C_1 R_{3(\min)}/R_{4(\max)} = 0.1 \mu\text{F} \times 1 \text{ k}\Omega/200 \text{ k}\Omega = 500 \text{ pF} \end{aligned}$$

$$\begin{aligned} 10\text{-}2 \quad C_5 &= C_1 R_3/R_4 = 0.1 \mu\text{F} \times 1 \text{ k}\Omega/2.1 \text{ k}\Omega = 0.048 \mu\text{F} \\ R_5 &= R_1 R_4/R_3 = 109.5 \Omega \times 2.1 \text{ k}\Omega/1 \text{ k}\Omega = 230 \Omega \\ D &= \omega C_5 R_5 = 2 \pi \times 1 \text{ kHz} \times 0.048 \mu\text{F} \times 230 \Omega = 0.069 \end{aligned}$$

$$10\text{-}3 \quad R_5 = (R_1 R_4/R_3) - 3.9 \Omega$$

Solution Manual For Electronic Instrumentation Measurements

Northrop Robert B, Northrop/Robert B.



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CRC Handbook of Basic Tables for Chemical Analysis Thomas J. Bruno, Paris D.N. Svoronos, 2020-07-30 Researchers in chemistry chemical engineering pharmaceutical science forensics and environmental science make routine use of chemical analysis but the information these researchers need is often scattered in different sources and difficult to access The CRC Handbook of Basic Tables for Chemical Analysis Data Driven Methods and Interpretation Fourth Edition is a one stop reference that presents updated data in a handy format specifically designed for use when reaching a decision point in designing an analysis or interpreting results This new edition offers expanded coverage of calibration and uncertainty and continues to include the critical information scientists rely on to perform accurate analysis Enhancements to the Fourth Edition Compiles a huge array of useful and important data into a single convenient source Explanatory text provides context for data and guidelines on applications Coalesces information from several different fields Provides information on the most useful wet chemistry methods as well as instrumental techniques with an expanded discussion of laboratory safety Contains information of historical importance necessary to interpret the literature and understand current methodology Unmatched in its coverage of the range of information scientists need in the lab this resource will be referred to again and again by practitioners who need quick easy access to the data that forms the basis for experimentation and analysis

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