

Solubility Rules Worksheet

1. Classify each substance as being soluble or insoluble in water.

- a. Potassium bromide – _____ f. Silver iodide - _____
b. Lead (II) carbonate – _____ g. Cadmium (II) sulfide - _____
c. Barium sulfate – _____ h. Zinc carbonate - _____
d. Zinc hydroxide – _____ i. Silver acetate - _____
e. Sodium acetate – _____ j. Copper (II) sulfide – _____

2. Identify the two new compounds which form if the solutions were mixed. Circle any compounds that would precipitate from the solution (are insoluble).

	KBr	Na_2CO_3	CaS	NH_4OH
AgNO_3				
BaCl_2				
$\text{Al}(\text{NO}_3)_3$				
CuSO_4				

3. Name or give the chemical formula for each of the following compounds and state whether they are soluble (will dissolve) or insoluble (will not dissolve) in solution.

Chemical Formu	Name	Solubility
$\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$		
	Barium hydroxide	
FeCO_3		
NaOH		
	Rubidium nitrate	
MgSO_4		
	Zinc chloride	
	Zinc hydroxide	

Solubility Rules Lab Answer Key

**V. van Schijnde, K. Cutts, I. Pereira, M.
Guitreau, S. Volante, M. Tedeschi**

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General, Organic, and Biological Chemistry Study Guide and Selected Solutions Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems You Want Me to Teach What? Norman Joseph LaFave, 2012 Problem You feel shaky about being assigned to teach upper level science and math and need to get up to speed fast Solution Follow this concise book's tried and true methods which you can integrate into your classroom and lesson plans starting from the first day of class You Want Me to Teach What avoids long discussions of education theory and specific lesson plans Instead it concentrates on general techniques for approaching a variety of problems and enhancing your teaching skills in science and math It covers student psychology classroom management planning instruction problem solving techniques laboratory methods and reporting assessment and professional development Without feeling inundated you'll find a wealth of sensible guidance whether you're a preservice education major wanting to teach physical science or mathematics a new teacher looking for practical methods to integrate into your instruction or an experienced teacher in search of fresh ways to improve in the classroom *Teaching High School Science Through Inquiry and Argumentation* Douglas Llewellyn, 2013 For Grades 9-12 this new edition covers assessment questioning techniques to promote learning new approaches to traditional labs and activities that emphasize making claims and citing evidence

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generic structural elements including test planning test matrices sampling conditioning test procedure selection data reporting data reduction statistical analysis and other related topics Special attention is given to the statistical treatment and analysis of data Volume 1 contains guidelines for general development of material characterization data as well as specific requirements for publication of material data in CMH 17 The primary purpose of this volume of the handbook is to document industry best practices for engineering methodologies related to testing data reduction and reporting of property data for current and emerging composite materials It is used by engineers worldwide in designing and fabricating products made from composite materials The Composite Materials Handbook referred to by industry groups as CMH 17 is a six volume engineering reference tool that contains thousands of records of the latest test data for polymer matrix metal matrix ceramic matrix and structural sandwich composites CMH 17 provides information and guidance necessary to design analyze fabricate certify and support end items using composite materials It includes properties of composite materials that meet specific data requirements as well as guidelines for design analysis material selection manufacturing quality control and repair

Microscale and Miniscale Organic Chemistry Laboratory Experiments Allen M. Schoffstall, Barbara A. Gaddis, Melvin L. Druelinger, 2000 This work offers a comprehensive introductory treatment of the organic laboratory techniques for handling glassware and equipment safety in the laboratory micro and mini scale experimental procedures theory of reactions and techniques applications and spectroscopy Energy Research Abstracts, 1993 *The Office of Environmental Management Technical Reports*, 1997 **Process Systems and Materials for CO₂ Capture** Athanasios I. Papadopoulos, Panos Seferlis, 2017-03-07 This comprehensive volume brings together an extensive collection of systematic computer aided tools and methods developed in recent years for CO₂ capture applications and presents a structured and organized account of works from internationally acknowledged scientists and engineers through Modeling of materials and processes based on chemical and physical principles Design of materials and processes based on systematic optimization methods Utilization of advanced control and integration methods in process and plant wide operations The tools and methods described are illustrated through case studies on materials such as solvents adsorbents and membranes and on processes such as absorption desorption pressure and vacuum swing adsorption membranes oxycombustion solid looping etc **Process Systems and Materials for CO₂ Capture Modelling Design Control and Integration** should become the essential introductory resource for researchers and industrial practitioners in the field of CO₂ capture technology who wish to explore developments in computer aided tools and methods In addition it aims to introduce CO₂ capture technologies to process systems engineers working in the development of general computational tools and methods by highlighting opportunities for new developments to address the needs and challenges in CO₂ capture technologies **Scientific and Technical Aerospace Reports**, 1981 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database

Fusion Energy Update ,1983 Minor Minerals, Major Implications: Using Key Mineral Phases to Unravel the Formation and Evolution of Earth's Crust V. van Schijnde,K. Cutts,I. Pereira,M. Guitreau,S. Volante,M. Tedeschi,2024-01-11
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