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Chapter 5 The Periodic Table

Investigation SA

Using Clues to Identify Elements

Background Information

Chemical elements can be classified according to their properties as metals, normetals, and metalloids. Metals are good conductors of heat and electricity. Many metals are malliable and ductile. Nonmetals are poor conductors of heat and electricity, and solid normetals tend to be brittle. Metalloids have properties between those of metals and nonmetals.

Elements in the same group on the periodic table have the same number of valence electrons, which are electrons in the highest occupied energy level of an atom. The number of an A group matches the number of valence electrons in atoms of each element in the group. For example, the Group 4A elements each have four valence electrons per atom. The examption to this pattern is the element helium, which is in Group 8A but has only two valence electrons.

Because elements in a group have the same reamber of valence electrons, they tend to have similar properties. The most reactive metals are the alkali metals in Group IA on the far left side of the periodic table. The Group IA elements are the alkaline earth metals, which are somewhat less reactive than the alkali metals. Groups of elements become less metallic in their properties from left to right across the periodic table. The most reactive more from left to right across the periodic table. The most reactive more from the thich are colories and odorless, and narely react with other elements.

In this investigation, you will use a list of closes to identify 34 elements and place them in their correct locations in the periodic table.

Problem

Where do the elements described in the class fit in the periodic table?

Pre-Lab Discussion

Bend the entire investigation. Then, work with a partner to answer the following questions.

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Using Clues To Identify Elements Physical Science

IM Harris

Using Clues To Identify Elements Physical Science:

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

Challenging Puzzles-Physical Science Janet Aaker Smith,2009 Critical and creative reasoning puzzles can be used as curriculum extensions and as anchor activities in the differentiated classroom for pre and post testing or as an introduction to a new unit Puzzles can be completed by individuals or small groups placed in learning centers or used as a presentation to the entire class using teacher made transparencies Challenging Puzzles Physical Science includes critical and creative reasoning puzzles some of which require research

Physical Science, Grades 4 - 6 Armstrong, 2009-02-16 Connect students in grades 4 6 with science using Physical Science Daily Skill Builders This 96 page book features two short reproducible activities per page and includes enough lessons for an entire school year It covers topics such as simple machines and alternative energy sources understanding the behavior and uses of electricity and framing scientific questions and recognizing scientific evidence Activities allow for differentiated instruction and can be used as warm ups homework assignments and extra practice The book supports National Geography Harcourt Science: Physical science, [grade] 5, Units E and F, teacher's ed ,2000 Standards Physical Science Robert P. Bauman, 1987 This text provides an understanding of the foundations and structure of physical science by emphasizing science as a search for truth rather than an accumulation of facts It develops the subject through concrete examples such as inclined planes and levers speeds and displacements progressing to consideration of forces and the concept of inertia and the idea of energy Similarly a study of observable chemical reactions advances to the ways in which atoms combine separate and displace one another and how observable masses of reactants and products illustrate how the atoms are combining The fundamental ideas are applied to astronomy optics geology music and the chemical compounds of life processes. The wide variety of end of chapter problems and multiple choice questions reinforce comprehension of each Introduction to Concepts and Theories in Physical Science Gerald James Holton, 1962 **Astronomy and** topic Astrophysics for the 1980's, Volume 1 National Research Council, Division on Engineering and Physical Sciences, Commission on Physical Sciences, Mathematics, and Applications, Astronomy Survey Committee, 1982-02-01 An **Approach to Physical Science** Physical Science for Nonscience Students Project, 1969 **Glencoe Physical Science Reading with Meaning** Dorothy Grant Hennings, 2005 McGraw Hill, 1999 2000 2005 State Textbook Adoption How Societies Work, 5th Edition Joanne Naiman, 2020-04-28T00:00:00Z In 2011 protesters around the world including Canada called for changes to the societies in which they live Many observers were asking What do they want Some answers to this

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