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web hclo 4 is the stronger acid because in a series of oxyacids with similar formulas the higher the electronegativity of the central atom the stronger is the attraction of the central atom for the electrons of the oxygen s the stronger attraction of the oxygen electron results in a stronger attraction of oxygen for the electrons in the o h bond making the hydrogen

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web hexane is a nonpolar liquid with a dipole moment of zero and therefore does not significantly interact with the ions of the nacl crystals 11 a fe no 3 3 is a strong electrolyte thus it should completely dissociate into fe 3 and no 3 ions therefore z best represents the solution b fe no 3 3 s fe 3 a q 3 no 3 a q

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following questions in the space provided 1 match the solution type on the right to the corresponding relationship between the ion product and the K_{sp} for that solution listed on the left the ion product exceeds

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web here the value of concentration of water is taken as 0 36 instead of 0 16 divide the product of the concentration of water carbon monoxide and the equilibrium constant by the concentration of hydrogen instead of multiplying it correct d the value of K_{eq} is taken as 4 4 instead of 0 44