

FINANCIAL ACCOUNTING

VOLUME TWO

Valix Peralta Valix

2013 Edition

SOLUTION MANUAL

1

CHAPTER 1

Problem 1-1

1. C
2. D
3. D
4. C
5. A

Problem 1-2

1. A
2. D
3. B
4. C
5. D

Problem 1-3

1. A
2. D
3. C
4. A
5. C

Problem 1-4

- | | |
|------|-------|
| 1. A | 6. C |
| 2. A | 7. A |
| 3. B | 8. C |
| 4. A | 9. C |
| 5. B | 10. A |

Problem 1-5

1. C
2. D
3. A
4. A
5. C

Problem 1-6

1. D
2. A
3. C
4. B
5. C

Problem 1-7

Accounts payable	1,000,000
Deposits and advances from customers	250,000
Notes payable	1,000,000
Credit balances in customers' accounts	200,000
Serial bonds payable	1,000,000
Accrued interest on bonds payable	150,000
Unearned rent income	<u>100,000</u>
Total current liabilities	<u>3,700,000</u>

Problem 1-8

Note payable – trade	3,000,000
Note payable – bank	2,000,000
Note payable – officers	500,000
Accounts payable – trade	4,000,000
Bank overdraft	300,000
Dividends payable	1,000,000
Withholding tax payable	100,000
Income tax payable	800,000
Estimated income tax liabilities	<u>600,000</u>

Solution Manual Financial Accounting By Valix

KJ Lindholm-Leary



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web 1 sodium hydroxide solution is poured into a beaker of hydrochloric acid which contains a thermometer showing room temperature 2 the beaker now contains sodium chloride and water and the

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable measure an appropriate volume of each liquid eg 25 cm³ 3 place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with HCl as 57.9 kJ mol⁻¹

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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic eg 1 HCl(aq) + NaOH(aq) → NaCl(aq) + H₂O(l) for which ΔH = 57 kJ mol⁻¹ eg 2 HNO₃(aq) + KOH(aq) → KNO₃(aq) + H₂O(l) for which ΔH = 57 kJ mol⁻¹

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