

Solubility Webquest

Take the webquest questions answer the solubility questions

Part 1

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1. Explain a substance's general solubility and how it is related to its molecular weight.
2. What are the two parts of a solution's composition? Explain what is changing in it as solute and solvent interact and describe the energy changes that occur when higher concentrations of solute prevent the crystals from dissolving to further.
3. Explain what is meant by a solution's concentration and how it is related to its molecular weight.
4. Explain what is meant by the rate of a solution's composition.
5. Explain what is meant by a solution's composition and how it is related to its molecular weight.
6. Explain what is meant by the rate of a solution's composition.
7. Explain what is meant by the rate of a solution's composition and how it is related to its molecular weight.
8. Explain what is meant by a solution's composition and how it is related to its molecular weight.
9. Explain what is meant by a solution's composition and how it is related to its molecular weight.
10. Explain what is meant by a solution's composition and how it is related to its molecular weight.

Part 2

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1. The composition of a solution is a measure of the amount of solute in a given amount of solvent.
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Part 3

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Karin Nielsen-Saines



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web answer mark 5 b ii 1 question number answer mark 5 c ii 1 question number answer mark 5 d iii 1 question number answer mark 6 a d e h 1 mark for every correct

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web apr 19 2016 assessment weightings the marks awarded on the papers will be scaled to meet the weighting of the components students final marks will be calculated by adding

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