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## Solubility Study Guide

Name	
Teacher:	
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#### References

Chemistry in the Community pp 45-65 Water properties concept map Solubility concept map Solubility, graphing and analysis Solubility Practice: A Curvy Subject Study Solvent Lab

#### Vocabulary

Understand and apply the definitions or symbols of the following terms.

Colligative properties Distillation Filtration Homogeneous Mixtures Nonpolar Polar Pure substances Solubility Solute
Solvent
Solution
Supersaturated
Saturated
Unsaturated
ö+ and ö-

#### Targeted Skills

- Describe the structure of water and relate its unusual physical properties to its structure, high boliling point, surface tension, high specific heat.
- Draw the structure of water and relate its structure to the ability of water to act as a universal solvent.
- 3. Compare/contrast the solubilities of various ionic and covalent solutes.
- 4. Apply: Like dissolves like:

Polar solvents dissolve Polar and Ionic solutes

Nonpolar solvents dissolve Nonpolar solutes

- Describe factors that affect the rate of dissolving, such as particle size, stirring, temperature, nature of the solid solutes.
- Interpret information from a solubility curve (graph).

#### Make sure you remember from past tests?

Atomic structure Significant figures Accuracy & precision

#### Equations:(extension)

Solution concentration = amount of solute

amount solute + solvent

#### Test Breakdown

30 Multiple-Choice Problems (max.) - 60%

1 Free Response Problem chosen from the Free Response Test Bank - 10%

1 Open-Ended Problem chosen from the Open-Ended Test Bank - 10%.

1 Performance Assessment - 20%

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**S Marginson** 

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