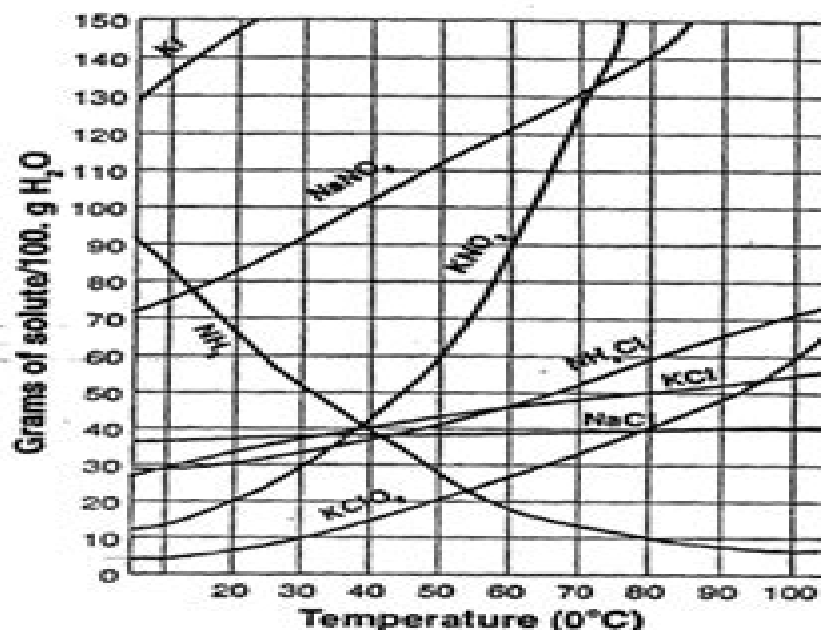


SOLUBILITY CURVES

Name _____

Answer the following questions based on the solubility curve below.

- Which salt is least soluble in water at 20° C? KClO₃
- How many grams of potassium chloride can be dissolved in 200 g of water at 80° C?
100 g
- At 40° C, how much potassium nitrate can be dissolved in 300 g of water? 123 g
- Which salt shows the least change in solubility from 0° - 100° C?
NaCl
- At 30° C, 90 g of sodium nitrate is dissolved in 100 g of water. Is this solution saturated, unsaturated or supersaturated?
unsaturated
- A saturated solution of potassium chlorate is formed from one hundred grams of water. If the saturated solution is cooled from 80° C to 50° C, how many grams of precipitate are formed? 40 - 20 g = 20 g
- What compound shows a decrease in solubility from 0° to 100° C? NH₃
- Which salt is most soluble at 10° C? KI
- Which salt is least soluble at 50° C? KClO₃
- Which salt is least soluble at 90° C? NH₃



Solubility Curves Practice 1 Answer

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A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the Kaplan Test Prep text.

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