# S'more Stoichiometry





**Definitions:** 'Stoichio' means element and 'metry' means the process of measuring. The mass and quantity relationships among reactants and products in a reaction are found using the process of stoichiometry.

Problem:

- If you are given one bag of large marshmallows, what is the maximum number of S'mores that can be made?
- 2) How many baxes of graham crackers and how many chocolate bars are needed to make this many \$'mores?

Solutions:

1st Step: Write a chemical equation using the following symbols:

Substance	Symbol	Unit Mass
Graham Cracker	S	7.00 g
Marshmallow	Mm	7.10 g
Chocolate Pieces	Or	3.30 g
S'more	SoMmOra	a

Calculate the unit mass of the S'more (S2MmOr3)below:

2<sup>nd</sup> Step: Balance the equation: What does the equation tell us? What do the coefficients represent?

They represent the ratio of the \_\_\_\_\_ or the \_\_\_\_\_.

# **Smores Stoichiometry Lab**

**Marco Cascella** 

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