

Total No. of Questions : 6]

SEAT No. :

P462

[Total No. of Pages : 3

[5115] - 205
S.Y. B.Sc. (Semester - II)
CHEMISTRY
CH - 221: Physical and Analytical Chemistry
(2013 Pattern)

Time : 2 Hours]

[Max. Marks : 40

Instructions to the candidates:

- 1) All questions are compulsory.
- 2) Neat diagram must be drawn wherever necessary.
- 3) Figures to the right side indicate full marks.
- 4) Use of calculator is allowed.
- 5) Answer to both sections should be written in separate answer sheet.

SECTION - I
[Physical Chemistry]

Q1) Answer the following. [5]

- a) State the criteria of equilibrium in terms of ΔF .
- b) Write the Gibbs - Helmholtz equation.
- c) Draw the T-N diagram of solution with negative deviation.
- d) Define the term: Lower critical solution temperature.
- e) Define the term: Normality.

Q2) a) Attempt any two of the following. [6]

- i) Derive the Clapeyron equation.
- ii) State and explain Henry's law. Give its applications.
- iii) Explain with diagram the mutual solubility behaviour of phenol - water system at various temperatures.

b) Answer any two of the following. [4]

- i) State Raoult's law and give its equation.
- ii) Explain the physical significance of ΔF .
- iii) How will you calculate graphically the heat of vaporisation of liquid from Clausius - Clapeyron equation.

P.T.O.

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**Gurumeet C. Wadhawa, 1st, Vitthal S.
Shivankar, 2nd, Supriya Pratap Babar,
3rd, Marayappa Sonawale, IV**

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Chemistry Practical for Undergraduate and Postgraduate Students Gurumeet C. Wadhawa, 1st, Vitthal S. Shivankar, 2nd, Supriya Pratap Babar, 3rd, Marayappa Sonawale, IV, 2022-03-16 We are very pleased and feel honored to present the First Edition of M Sc II Chemistry Practicals Book in the hands of our students of Mumbai University from the academic year 2021 22 All the experiments are written as prescribed by University's Syllabus Committee We have taken extreme care to keep all the experiments easy and simple All the experiments are rearranged as per the University Examination pattern We sincerely hope that the colleagues from the teaching faculty as well as students in particular will adapt to all the experiments described in this edition

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