

The Soviet Union was a communist giant. It started after the Russian Revolution in the early 1920s and lasted until 1991. But what should a textbook writer say about it today? Three things should be emphasized: its cultural achievements, its military strength, and its political oppression.

The first area that textbook authors should emphasize is cultural achievements. This includes Russia's Olympic success in getting the most gold medals between 1952 and 1988 (Doc G) but also its incredible ballet (Doc H). It recruited children all across the country and became world class. The Soviet Union was about a lot more things than bombs and prisons.

A second emphasis should be the military. The Soviets had a lower GNP and per capita GNP than the US (Doc B) yet their military spending in 1980 was \$165 billion to only \$131 billion for the US (Doc E). They passed the US in their number of nuclear weapons and submarines (Doc E). This made the Soviets a superpower and history should never ignore a superpower.

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DIEGO PIZANO was

economic advisor to the President of Colombia between 1982 and 1986 He is the Chairman of the Board of the University of the Andes Bogot Colombia and the President of the Common Code for the Coffee Community Association Bonn Germany

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web an solution initially contains a catalytic amount of an enzyme with $k_m = 1.5 \text{ mmol/L}$ of substrate and no product after 45 seconds the solution contains $25 \mu\text{M}$ of product find v_{max} and the concentration of product after 20 minutes hint $s = k_m$
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web a reaction and its experimental rate law are represented below $2 \text{ICl} + \text{H}_2 \text{g} \rightarrow 2 \text{HCl} + \text{I}_2 \text{g}$ rate $k [\text{ICl}]^2 [\text{H}_2]$ a chemist proposes a mechanism for the reaction that is consistent with the rate law the mechanism has two elementary steps and the first step is slow compared to the second

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web solution a first order with respect to Br_2 first order with respect to BrO_3^- and second order with respect to H^+ hence the overall order of the reaction is equal to $1 + 1 + 2 = 4$ b order of the reaction with respect to acetaldehyde is $3/2$ and overall order is also $3/2$

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nozzle is clogged there is an interference in the flow of gas which affects the performance of the burner which of the following best describes this scenario and the effect on the kinetics 2 practice problem

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worked example determining a rate law using initial rates data first order reaction with calculus plotting data for a first order reaction half life of a first order reaction

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web dec 26 2015 in the nuclear industry workers use a rule of thumb that the radioactivity from any sample will be relatively harmless after 10 half lives calculate the fraction of a radioactive sample that remains after this time period hint radioactive decay obeys first

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web in order to have successful collisions the colliding particles must have both the proper amount of energy and the proper fill in blank draw an energy diagram which shows the shape of the curve you might expect for the reaction in this question the overall reaction is exothermic

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web mar 13 2023 work in groups on these problems you should try to answer the questions without referring to your textbook if you get stuck try asking another group for help 1 for the reaction below if substance a is disappearing at a rate of $1.82 \times 10^{-2} \text{ mol l}^{-1} \text{ s}^{-1}$ at what rate is c appearing 3 a 3 b $\rightarrow$ 5 c 2 d

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consider the following mechanism
 $\text{O}_3 \rightarrow \text{O}_2 + \text{O}$ fast
 $\text{O}_3 + \text{O} \rightarrow 2 \text{O}_2$ slow
a) write the overall balanced chemical equation
b) identify any intermediates within the mechanism

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 $\text{Mg} + 2 \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$
g the reaction between Mg and HCl is represented by the equation above in a kinetics experiment
a) 0.080 g sample of Mg ribbon is added to an Erlenmeyer flask containing 50 mL of 3.0 M HCl at 25 °C

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studying chemical kinetics determines whether to proceed with a reaction as it measures the rate of a reaction
reactions conducted in an industrial facility mix compounds together heating and stirring them for a