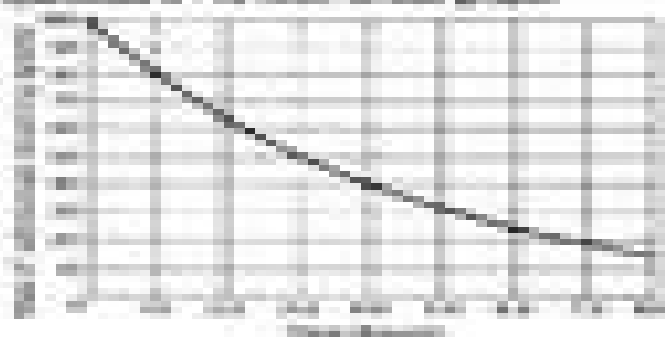


2-4 Nuclear Chemistry

AP CHEMISTRY TEST

2. When Strontium-90 decays, it emits a beta particle. Which of the following is the most "radioactive" Sr-90 emitting radionuclide?
- (A) having the longest half-life
 (B) having the most "radioactive" Sr-90 emitting radionuclide
 (C) having the shortest half-life
 (D) having the most Sr-90 emitting radionuclide
3. A sample of a substance with a half-life of 10 years is found to contain 25% of the original amount. How long has the sample been decaying?
- (A) 10 years
 (B) 20 years
 (C) 30 years
 (D) 40 years
4. A sample of a substance with a half-life of 10 years is found to contain 12.5% of the original amount. How long has the sample been decaying?
- (A) 10 years
 (B) 20 years
 (C) 30 years
 (D) 40 years
5. A sample of a substance with a half-life of 10 years is found to contain 6.25% of the original amount. How long has the sample been decaying?
- (A) 10 years
 (B) 20 years
 (C) 30 years
 (D) 40 years
6. A sample of a substance with a half-life of 10 years is found to contain 3.125% of the original amount. How long has the sample been decaying?
- (A) 10 years
 (B) 20 years
 (C) 30 years
 (D) 40 years
7. A sample of a substance with a half-life of 10 years is found to contain 1.5625% of the original amount. How long has the sample been decaying?
- (A) 10 years
 (B) 20 years
 (C) 30 years
 (D) 40 years

Question 8 - Refer to the graph.



8. According to the above data, what is the half-life of the substance?
- (A) 20 years
 (B) 40 years
 (C) 60 years
 (D) 80 years
9. What percent of the original sample remains after 80 years?
- (A) 25%
 (B) 31.64%
 (C) 42.19%
 (D) 56.25%
10. What is the half-life of the substance shown in the graph above?
- (A) 20 years
 (B) 40 years
 (C) 60 years
 (D) 80 years
11. A sample of a substance with a half-life of 10 years is found to contain 12.5% of the original amount. How long has the sample been decaying?
- (A) 10 years
 (B) 20 years
 (C) 30 years
 (D) 40 years
12. Which of the following describes what occurs in the reaction process?
- (A) a heavy nucleus is fragmented into lighter ions
 (B) a nucleus is split into a proton and an electron
 (C) two light nuclei are combined into a heavier one
 (D) a particle and an antiparticle annihilate completely into energy

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D Kirk



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