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# Valence Electrons

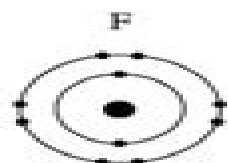
1 How many electrons are present in the following elements?

- a Carbon \_\_\_\_ b Chlorine \_\_\_\_ c Indium \_\_\_\_ d Calcium \_\_\_\_  
e Tungsten \_\_\_\_ f Antimony \_\_\_\_ g Titanium \_\_\_\_ h Tin \_\_\_\_

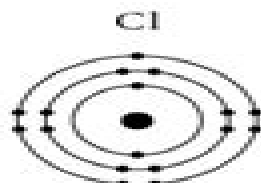
2 Using the periodic table, determine the number of valance electrons for the elements listed below.

- a Copper \_\_\_\_ b Xenon \_\_\_\_ c Hydrogen \_\_\_\_ d Aluminum \_\_\_\_  
e Barium \_\_\_\_ f Iodine \_\_\_\_ g Carbon \_\_\_\_ h Phosphorous \_\_\_\_

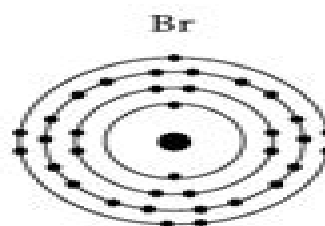
3 Circle the valence electrons only. What is the atomic number?



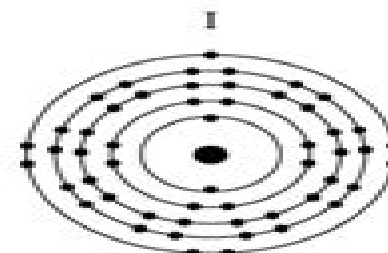
Atomic number(Z) =



Atomic number(Z) =



Atomic number(Z) =



Atomic number(Z) =

4 Fill in the table with correct values.

| Element   | Ion symbol     | Number of Protons | Number of electrons | Charge | Ion type |
|-----------|----------------|-------------------|---------------------|--------|----------|
| Fluorine  | F <sup>-</sup> |                   |                     | -1     | Anion    |
|           |                | 53                | 54                  |        |          |
|           |                | 16                |                     | -2     |          |
| Potassium |                |                   |                     | +1     |          |
| Calcium   |                |                   |                     |        |          |

# Valence Electrons Answer Key

**Mocktime Publication**

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