

12

STOICHIOMETRY

SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353–358)

This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

► Using Everyday Equations (pages 353–355)

1. How can you determine the quantities of reactants and products in a chemical reaction?

2. Quantity usually means the _____ of a substance expressed in grams or moles.
3. A bookcase is to be built from 3 shelves (Sh), 2 side boards (Sb), 1 top (T), 1 base (B), and 4 legs (L). Write a “balanced equation” for the construction of this bookcase.

► Using Balanced Chemical Equations (page 354)

4. Is the following sentence true or false? Stoichiometry is the calculation of quantities in chemical reactions. _____
5. Calculations using balanced equations are called _____.

► Interpreting Chemical Equations (pages 356–357)

6. From what elements is ammonia produced? How is it used?

7. Circle the letter of the term that tells what kind of information you CANNOT get from a chemical equation.

a. moles

d. volume

b. mass

e. number of particles

c. size of particles

Stoichiometry Section 9 1 The Arithmetic Of Equations Answers

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General College Chemistry Charles William Keenan, Donald C. Kleinfelter, Jesse Hermon Wood, 1980 *Addison-Wesley Chemistry* Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L. Waterman, Prentice-Hall Staff, 2001-02 To purchase or download a workbook click on the Purchase or Download button to the left To purchase a workbook enter the desired quantity and click Add to Cart To download a free workbook right click the FREE Download PDF link and save to your computer This will result in a faster download as opposed to left clicking and opening the link **Computers & Electronics**, 1985 Resources in Education, 1980 **Instructor's Manual, General Chemistry, Second Edition** Becker, Ralph Sherman Becker, 1980 The Software Finder, 1983 *Chemistry* David W. Oxtoby, Wade A. Freeman, Toby F. Block, 1998 Textbook of chemistry concepts Chemistry, an Experimental Science Chemical Education Material Study, 1963 **The Computing Teacher**, 1979 The Software Encyclopedia, 1986 **General Chemistry** James E. Brady, Gerard E. Humiston, 1980 **The Software Encyclopedia 2000** Bowker Editorial Staff, 2000-05

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