STOICHIOMETRY: LIMITING REAGENT

Name _____

- N₂ + 3H₂ → 2NH₃
 How many grams of NH₃ can be produced from the reaction of 28 g of N₂ and 25 g of H₂?
- How much of the excess reagent in Problem 1 is left over?
- Mg + 2HCl → MgCl₂ + H₂
 What volume of hydrogen at STP is produced from the reaction of 50.0 g of Mg and the equivalent of 75 g of HCl?
- 4. How much of the excess reagent in Problem 3 is left over?
- 3AgNO₃ + Na₃PO₄ → Ag₃PO₄ + 3NaNO₃
 Silver nitrate and sodium phosphate are reacted in equal amounts of 200. g each. How many grams of silver phosphate are produced?
- How much of the excess reagent in Problem 5 is left?

Stoichiometry Limitimg Reagent Pg 66 Answers

KJ Lindholm-Leary

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