Chapter Test B

Chapter: Stoichiometry

PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- Knowing the mole ratio of a reactant and product in a chemical reaction would allow you to determine
 - a. the energy released in the reaction.
 - b. the speed of the reaction.
 - c. the mass of the product produced from a known mass of reactant.
 - d. whether the reaction is reversible.
 - 2. In the equation 6CO₂ + 6H₂O → C₆H₃₂O₆ + 6O₂, the mole ratio of water to copygen is
 - a. 2:1.
 - Bs. 1:1.
 - c. 3:2.
 - d. 6:8.
- 3. In the equation Al₂(SO₄)₂ + 3Ca(OH)₂ → 3CaSO₄ + 2Al(OH)₃, the mole ratio of calcium hydroxide to aluminum hydroxide is
 - m. 1:3.
 - b. 2:3.
 - c. 1:1.
 - d. 3:2.
- 4. For the reaction represented by the equation N₂ + 3H₂ → 2NH₃, how many moles of nitrogen are required to produce 18 mol of ammonia?
 - a. 9
 - b. 18
 - c. 27
 - di, 36
- 5. In the reaction represented by the equation A + B → C + D, if there is more reactant B than is required to completely react with all of A, then
 - a. A is the limiting reactant.
 - b. B is the limiting reactant.
 - c. there is no limiting reactant.
 - d. no product can be formed.

Stoichiometry Chapter Test B

RD Boyd

Stoichiometry Chapter Test B:

Reviewing Stoichiometry Chapter Test B: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Stoichiometry Chapter Test B**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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