

Post-Lab Questions

1. Based on your X-Y graph, which one of Na_2CO_3 or H_2SO_4 is the limiting reactant in each of the seven flasks in the experiments?

Flask#	1	2	3	4	5	6	7
Limiting Reactant							

2. Based on your experimental results, what is the molar ratio of CO_2 (a product) and Na_2CO_3 (a reactant)?

3. If you react 1.205 g of Na_2CO_3 with 1.205 g of H_2SO_4 , how many grams of CO_2 will be produced.

Stoichiometry By Loss Of Co₂ Lab Answers

Roman Wölfel



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