

Working a Stoichiometry Problem

6.50 grams of aluminum reacts with an excess of oxygen. How many grams of aluminum oxide are formed?



$$6.50 \times 2 \times 101.96 \div 26.98 \div 4 = 12.3 \text{ g Al}_2\text{O}_3$$

6.50 g Al	1 mol Al	2 mol Al₂O₃	101.96 g Al₂O₃
	26.98 g Al	4 mol Al	1 mol Al₂O₃

$$= ? \text{ g Al}_2\text{O}_3$$

Stoichiometry Problem Solving Packet Holt Chemistry

**Rod O'Connor, Charles Mickey, Alton
Hassell**



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Holt Chemistry Ralph Thomas Myers,2004 **Books in Print Supplement** ,2002 **Children's Books in Print, 2007** ,2006 *Children's Books in Print* R R Bowker Publishing,Bowker,1999-12 **Holt Chemistry File** ,1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket **The Software Encyclopedia** ,1988 Energy Research Abstracts ,1993 **Holt Chemistry** ,2003 **Holt Chemistry 2** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2003-01-01 **Holt ChemFile Lab Program** ,2005 *Using the TAPPS Method to Increase the Problem Solving Skills of Chemistry Students During a Stoichiometry Unit* Dawn Toth,2004

Solving Problems in Chemistry R. O'Connor,C. Mickey,1974 Stoichiometry Michell J. Sienko,1964 **Solving Problems in Chemistry** Rod O'Connor,1977 *Stoichiometry and Structure* Michell J. Sienko,1964 *Stoichiometry and Structure. Freshman Chemistry Problems and how to Solve Them* Michell Joseph Sienko,1964 **Stoichiometry and Structure - Freshman Chemistry Problems and how to Solve Them - Part 1** ,1964 **Solving Problems in Chemistry** Rod O'Connor,Charles Mickey,Alton Hassell,1981 **Selected Solutions for Chemical Principles** Kenneth C. Brooks,Steven S. Zumdahl,1992 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples *Supporting Traditional Methods with a Constructivist Approach to Learning* Kenneth Munoz Abayan,ProQuest Dissertations and Theses (Electronic resource collection),2014 Stoichiometry is a fundamental topic in chemistry that measures a quantifiable relationship between atoms molecules etc Stoichiometry is usually taught using expository teaching methods Students are passively given information in the hopes they will retain the transmission of information to be able to solve stoichiometry problems masterfully Cognitive science research has shown that this kind of instructional teaching method is not very effecting in meaningful learning practice Instead students must take ownership of their learning The students need to actively construct their own knowledge by receiving interpreting integrating and reorganizing that information into their own mental schemas In the absence of active learning practices tools must be created in such a way to be able to scaffold difficult problems by encoding opportunities necessary to make the construction of knowledge memorable thereby creating a usable knowledge base Using an online e learning tool and its potential to create a dynamic and interactive learning environment may facilitate the learning of stoichiometry The study entailed requests from volunteer students IRB consent form a baseline questionnaire random assignment of treatment pre and post test assessment and post assessment survey These activities were given online A stoichiometry based assessment was given in a proctored examination at the University of Texas at Arlington UTA campus The volunteer students who took part in these studies were at least 18 of age and were enrolled in General Chemistry 1441

at the University of Texas at Arlington Each participant gave their informed consent to use their data in the following study Students were randomly assigned to one of 4 treatments groups based on teaching methodology Dimensional Analysis Operational Method Ratios and Proportions and a control group who just received instruction through lecture only In this study an e learning tool was created to demonstrate several methodologies on how to solve stoichiometry which are all supported by chemical education research Comparisons of student performance based on pre and post test assessment and a stoichiometry based examination was done to determine if the information provided within the e learning tool yielded greater learning outcomes compared to the students in the absence of scaffold learning material The e learning tool was created to help scaffold the problem solving process necessary to help students N 394 solve stoichiometry problems Therein the study investigated possible predictors for success on a stoichiometry based examination students conceptual understanding of solving stoichiometry problems and their explanation of reasoning It was found that the way the student answered a given stoichiometry question i e whether the student used dimensional analysis operational method or any other process was not statistically relevant $p > 0.05$ More importantly if the students were able to describe their thought process clearly these students scored significantly higher on stoichiometry test mean 84 p

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