## Intro to Stoichiometry - Moles to Moles NOTES

Reminder: To switch between moles of one substance and moles of another, use the ratio of the substances, which is found by looking at their coefficients.

 $N_2 + 3H_2 \rightarrow 2NH_3$ 



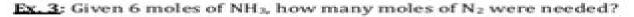


Answer: 6.0 moles of NH<sub>3</sub>



Ex. 2: Given 0.5 moles of H<sub>2</sub>, how many moles of NH<sub>3</sub> would be produced?

Answer: 0.33 moles of NH<sub>3</sub>



Answer: 3.0 moles of N2

Ex. 4: Given 1.5 moles of N2, how many moles of H2 were needed?

Answer: 4.5 moles of H2

# **Stoichiometry Mole Mole Problems Chemistry If8766**

**John Lambert** 

#### **Stoichiometry Mole Mole Problems Chemistry If8766:**

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