

Stoichiometry - Mixed Conversions Answer Key

1. Use the equation below the answer the following questions:



c. How many grams of NH_3 are made from 31L of N_2 gas?

$$\frac{31\text{L N}_2}{1} \times \frac{1\text{mol N}_2}{22.4\text{L N}_2} \times \frac{2\text{mol NH}_3}{1\text{mol N}_2} \times \frac{17\text{g/mol NH}_3}{1\text{mol NH}_3} = 47.3\text{g NH}_3$$

e. How many molecules of H_2 are needed to produce 66g of NH_3 ?

$$\frac{66\text{g NH}_3}{1} \times \frac{1\text{mol NH}_3}{17\text{g/mol NH}_3} \times \frac{1\text{mol N}_2}{2\text{mol NH}_3} \times \frac{6.02 \times 10^{23}\text{ molecules N}_2}{1\text{mol N}_2} = 1.22 \times 10^{24}\text{ molecules N}_2$$

f. How many moles of N_2 are needed to completely react with 87L of H_2 gas?

$$\frac{87\text{L H}_2}{1} \times \frac{1\text{mol H}_2}{22.4\text{L H}_2} \times \frac{1\text{mol N}_2}{3\text{mol H}_2} = 1.26\text{mol N}_2$$

2. Butane reacts with oxygen to form water and carbon dioxide.



c. How many molecules of H_2O are produced with 20.1L of O_2 gas that are completely used during the combustion of butane?

$$\frac{20.1\text{L O}_2}{1} \times \frac{1\text{mol O}_2}{22.4\text{L O}_2} \times \frac{10\text{mol H}_2\text{O}}{13\text{mol O}_2} \times \frac{6.02 \times 10^{23}\text{ molecules H}_2\text{O}}{1\text{mol H}_2\text{O}} = 4.17 \times 10^{23}\text{ molecules H}_2\text{O}$$

d. If 48g of H_2O are collected from the burning of butane, then how many moles of butane were reacted with oxygen?

$$\frac{48\text{g H}_2\text{O}}{1} \times \frac{1\text{mol H}_2\text{O}}{18\text{g H}_2\text{O}} \times \frac{2\text{mol C}_4\text{H}_{10}}{10\text{mol H}_2\text{O}} = 0.48\text{mol C}_4\text{H}_{10}$$

3. Potassium reacts with water to produce potassium hydroxide and hydrogen gas.



c. How many grams of H_2 will be produced if 7.22×10^{23} atoms of potassium react completely with water?

$$\frac{7.22 \times 10^{23}\text{ atoms K}}{1} \times \frac{1\text{mol K}}{6.02 \times 10^{23}\text{ atoms K}} \times \frac{1\text{mol H}_2}{2\text{mol K}} \times \frac{2\text{g H}_2}{1\text{mol H}_2} = 0.20\text{g H}_2$$

d. How many atoms of K are needed to completely react with 3.22×10^{23} molecules of water?

$$\frac{3.22 \times 10^{23}\text{ molecules H}_2\text{O}}{1} \times \frac{1\text{mol H}_2\text{O}}{6.02 \times 10^{23}\text{ molecules H}_2\text{O}} \times \frac{2\text{mol K}}{1\text{mol H}_2\text{O}} \times \frac{6.02 \times 10^{23}\text{ atoms K}}{1\text{mol K}} = 3.22 \times 10^{23}\text{ atoms K}$$

Stoichiometry Mixed Problems Key

Ying-Ying Zheng

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Stoichiometry Mixed Problems Key:

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Methods in Ecosystem Science Osvaldo E. Sala, Robert B. Jackson, Harold A. Mooney, Robert Howarth, 2000-08-11 Ecology at the ecosystem level has both necessitated and benefited from new methods and technologies as well as those adapted from other disciplines With the ascendancy of ecosystem science and management the need has arisen for a comprehensive treatment of techniques used in this rapidly growing field Methods in Ecosystem Science answers that need by synthesizing the advantages disadvantages and tradeoffs associated with the most commonly used techniques in both aquatic and terrestrial research The book is divided into sections addressing carbon and energy dynamics nutrient and water dynamics manipulative ecosystem experiments and tools to synthesize our understanding of ecosystems Detailed information about various methods will help researchers choose the most appropriate methods for their particular studies Prominent scientists discuss how tools from a variety of disciplines can be used in ecosystem science at different scales NBS Special Publication , 1968

Comprehensive Nuclear Materials , 2020-07-22 Materials in a nuclear environment are exposed to extreme conditions of radiation temperature and or corrosion and in many cases the combination of these makes the material behavior very different from conventional materials This is evident for the four major technological challenges the nuclear technology domain is facing currently i long term operation of existing Generation II nuclear power plants ii the design of the next generation reactors Generation IV iii the construction of the ITER fusion reactor in Cadarache France iv and the intermediate and final disposal of nuclear waste In order to address these challenges engineers and designers need to know the properties of a wide variety of materials under these conditions and to understand the underlying processes affecting changes in their behavior in order to assess their performance and to determine the limits of operation Comprehensive Nuclear Materials Second Edition Seven Volume Set provides broad ranging validated summaries of all the major topics in the field of nuclear material research for fission as well as fusion reactor systems Attention is given to the fundamental scientific aspects of nuclear materials fuel and structural materials for fission reactors waste materials and materials for fusion reactors The articles are written at a level that allows undergraduate students to understand the material while providing active

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Essentials of Chemical Reaction Engineering H. Scott Fogler, 2011 Accompanying DVD ROM contains many realistic interactive simulations **Scientific and Technical Aerospace Reports** ,1985

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