

Stoichiometric Calculations

Equation:	$2 \text{H}_2(\text{g})$	+	$\text{O}_2(\text{g})$	$\longrightarrow$	$2 \text{H}_2\text{O}(\text{l})$
Molecules:	2 molecules H_2	+	1 molecule O_2	$\longrightarrow$	2 molecules H_2O
					
Mass (amu):	4.0 amu H_2	+	32.0 amu O_2	$\longrightarrow$	36.0 amu H_2O
Amount (mol):	2 mol H_2	+	1 mol O_2	$\longrightarrow$	2 mol H_2O
Mass (g):	4.0 g H_2	+	32.0 g O_2	$\longrightarrow$	36.0 g H_2O

The coefficients in the balanced equation give the ratio of *moles* of reactants and products

Stoichiometric Calculations Answers

J Spring



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web it was played on the national viewers and listeners awards 1984 below is the script of this sketch and some excerpts in realvideo excerpts from the thatcher script script prime minister ah good morning jim sir humphrey do come in and sit down how s your wife is she well jim hacker puzzled oh yes fine prime minister fine thank

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