

Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

- 18 mol O₂ 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?



- 21 mol O₂ 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?



- 45 mol H₂O 3. How many moles of water are produced from 22.5 moles of oxygen?



- 20 mol HCl 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.



- 21 mol Cl₂ 5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



Stoichiometry Workbook Answer Key

Puja Mehta

A decorative graphic element consisting of a light blue horizontal bar with a rounded right end, and a red-to-white gradient semi-circle overlapping the right side of the bar.

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