g HNO₃
$$\xrightarrow{1}$$
 mol HNO₃ $\xrightarrow{2}$ mol Ba(NO₃)₂ $\xrightarrow{3}$ m Ba(NO₃)₂

Mole Ratio

$$16.5 \text{ g HNO}_{3} \times \frac{1 \text{ mol HNO}_{3}}{63.0 \text{ g HNO}_{3}} \times \frac{1 \text{ mol Ba(NO}_{3})_{2}}{2 \text{ mol HNO}_{3}} \times \frac{261.3 \text{ g}}{1 \text{ mol Ba(NO}_{3})_{2}} = 34.2 \text{ g}$$

use the molar mass of HNO₃ to calculate its moles

use mole ratio of

HNO₃ and

Ba(NO₃)₂ to

calculate the moles

of Ba(NO₃)₂

use moles of

Ba(NO₃)₂ to

calculate its mass

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